Synthesis of Some New Thioethers and 4-Thiazolidinones Bearing 3-(Pyridine-4’-yl)-1,2,4-Triazino[5,6-b]Indole Moiety as Antifungal Agents

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Abstract

Some new asymmetric thioethers 5 and 4-thiazolidinones 6 have been obtained from condensation of 5-formyl-3-(pyridin-4’-yl)-1,2,4-triazino[5,6-b]Indole (3) with halogenated aromatic amines followed by addition of thiophenol and/or cycloaddition with thiolactic acids in nonpolar solvents. Structures of the products confirmed by elemental analysis and spectral measurements. The new systems obtained were evaluated as antifungal agents.

Keywords

Thioethers, Thiazolidinones, 1,2,4-Triazino Indole, Fungicidal, Cycloaddition

1. Introduction

Heterocyclic systems of 1,2,4-Triazino[5,6-b] indole have been successfully used as a carrier for diverse functional groups in the development of some antiviral agents [1]. Various studies found the efficacy of such nucleus in the production of antibacterial [2], antifungal [3], inhibitors of blood platelet aggregation, anti-hypertensive [4] agents. As well as 1,2,4-triazino[5,6-b]indoles derivatives apply useful applications in medicinal chemistry, especially, antimalarial [5], and antidepressant agents [6]. Beside, 3,5-disubstituted-1,2,4-triazino[5,6-b]indoles exhibited an important biological activity as antileishmanial [7], antihypoxic [8] and anti-inflammatory activities [9]. On the other hand, recently found thiazolidinones showed a high activity towards the most fungi [10]. Based upon these observations, and in search for obtain a newly systems combined between thioether bearing 1,2,4-triazino[5,6-b]indole and thiazolidinones [11] [12].
metrical thioether bearing 1,2,4-triazine moiety exhibited anticancer and anti-HIV activity [10].

The present work tends to synthesize some new thiazolidinones bearing a triazinoindolone view of their effects and fungicidal activity.

2. Chemistry

A novel route to synthesize 3,5-disubstituted heterocyclic-1,2,4-triazino[5,6-b] indoles deduced from condensation of isatin with isonicotinic acid hydrazide via reflux in MeOH – drops AcOH to give 3-hydrazono-indol-2(1H)one (1) which upon fused with ammonia acetate in glacial AcOH, yielded 3-(pyridin-4'-yl)-5H-1,2,4-triazino[5,6-b]indole (Scheme 1). Careful formylation of compound 2 via stirring with formyl acetate (prepared by reflux equimolar amounts of HCO₂H with Ac₂O) in dry ether at room temperature, afforded 3-(pyridin-4'-yl)-5-formyl-1,2,4-triazino[5,6-b]indole (Scheme 1). Compound 3 underwent condensation with primary aromatic amines in EtOH yielded the Schiff bases 4 (Scheme 2). Simple 1,2-addition to -CH=N-Ar of compounds 4a-c under reflux with thiophenol in ethylbenzene to give the thioethers 5a-c (Scheme 2).

As well as cycloaddition of thiolactic acid to -CH=N-Arof compounds 4a-c, via reflux in dioxan, led to the direct formation of 4-thiazolidinones 6a-c, respectively, (Figure 1).

Figure 1. A plausible mechanism of 1,2-addition reaction of 4 to produce 6.

Scheme 1. Synthesis of compound 3.
3. Results and Discussion

Former structure of the synthesized compounds deduced from their correct elemental analysis and spectral data. IR spectrum of compound 1 showed ν at 3400 - 3300 and 3190 cm⁻¹ for OH and NH, while lacking of OH and C=O groups for 2. Compound 3 showed the presence of ν at 1738 cm⁻¹ for CHO formed and absence NH. IR spectra of compounds 4 and 6 recorded a lack of NH functional group.

On the other hand, IR spectra of 5 and 6 showed ν at 1200 - 1180 cm⁻¹ for C-S-C and ν at 1700 - 1680 cm⁻¹ for C=O of compounds 6 with ν at 2900, 2850, 1480 and 1440 cm⁻¹ attribute to aliphatic groups. All the new fluorinated compounds obtained showed ν at 1250 - 1240 cm⁻¹ for C-F, and ν at 1620 - 1590 cm⁻¹ for cyclic C=N groups of pyridine and 1,2,4-triazine.

¹HNMR spectra of the compounds 2-6 give us good indications about their line structures. Only compound 3 showed a resonated signal at 8.80 ppm for formyl proton, while that of compounds 2 and 5 recorded δ at 11.43 ppm attribute to NH protons. ¹HNMR spectra of compounds 4, 5 and 6 showed δ at 8.5, 7.8 - 7.7 and 2.52 ppm for pyridine, CH=N and CH₃ protons. All the 4-thiazolidinone derivatives 6a-c showed a resonated signal at δ 3.53 ppm attribute to C5-H.

¹³CNMR spectra of the compounds 3-6 showed δ at 162, 155, 145, 140, 130 - 120 and 77 - 44 ppm attribute to C=O, C-S-C, C-F, C=N, pyridine aromatic and aliphatic carbons.

In addition, mass spectrometric study of some compounds recorded the molecular ion peak at lower integration which upon further fragmentation process give a base peak at lower integration (Figure 1 and Figure 2). For example, M/z
of compound 3 recorded a base peak at 116 attribute to the indolyl ion, while that of 6a exhibited a base peak at 60. Stability of base peak for compound 6 may be due to a type of hyperconjugation bonded to thirene radical, a plausible Mass fragmentation patterns of compounds 3 and 6a are illustrated in (Figure 3 and Figure 4) respectively.

4. Antifungal Activity

4-Thiazolidinone derivatives exhibited a wide range of biological activity especially as antifungal agents [11] [13], also, thioethers bearing a heterocyclic nitrogen systems used as anticancer, and anti-HIV probes [10]. Besides, the 1,2,4-triazino[5.6- b] indole, showed various biological activities [4] [5] [6] [7]. Based on these observations, the present work tends to synthesize some new systems combined between these biocidal agents. The new compounds containing S, F, Cl and Br atoms (5 - 6) were evaluated as antifungal agents against Aspergillus niger and Alternaria alternata fungi, using the diffusion method [14] [15].

The tested compounds dissolved in DMF (1 mg·ml⁻¹) and the antibiotic Fluconazole used as a standard drug. The inhibition zones (microbial growth surrounding the filter paper disc (2.5 mm) were measured at the incubation period at 30°C for 3 days. The inhibition zones of the organisms measured by a zone surrounding each disk and reported in Table 1.

From the results obtained from the Table 1 we can conclude that:

1) The compounds containing F atoms are more active than that contain Cl and Br.

2) The more sulfur element percentage, showed higher inhibition zones towards fungi.

3) The compounds contains both F and S are very active than other compounds.
Figure 3. A suggested Mass fragmentation pattern of compound 3.

Figure 4. Mass fragmentation pattern of compound 6a.

Table 1. The fungicidal effects of some synthesized compounds.

<table>
<thead>
<tr>
<th>Compounds</th>
<th>% S</th>
<th>Aspergillus N.</th>
<th>Alternaria A.</th>
</tr>
</thead>
<tbody>
<tr>
<td>5a</td>
<td>6.61</td>
<td>29</td>
<td>26</td>
</tr>
<tr>
<td>5b</td>
<td>6.04</td>
<td>27</td>
<td>26</td>
</tr>
<tr>
<td>5c</td>
<td>5.93</td>
<td>25</td>
<td>25</td>
</tr>
<tr>
<td>6a</td>
<td>6.98</td>
<td>30</td>
<td>27</td>
</tr>
<tr>
<td>6b</td>
<td>6.28</td>
<td>28</td>
<td>26</td>
</tr>
<tr>
<td>6c</td>
<td>6.16</td>
<td>28</td>
<td>25</td>
</tr>
<tr>
<td>Fluconazole</td>
<td>-</td>
<td>25</td>
<td>25</td>
</tr>
</tbody>
</table>
4) The antifungal activities are in the order: \(6a > 5a > 6b, 6c > 5b > 5c\) for \(Aspergillus\ N\). and \(6a > 5a, 5b, 6b > 5c, 6c\) for \(Alternaria\ A\). fungi.

5) Most of the evaluated compounds showed IZ more than that of Fluconazole which makes these fluorine substituted sulfur systems good fungicidal probes.

5. Experimental

Melting points determined with an electrothermal Bibby Stuart Scientific melting point sample (UK). A Perkin Elmer Model PXI-FT system 55,529 used for recording IR spectra of the prepared compounds. A Bruker advance DPX 400 MHz model uses TMS as an internal standard was used for recording the \(^1\)H and \(^{13}\)CNMR spectra of the compounds on deuterated DMSO-\(d_6\). A GC-MS-GP 1000 Ex model used for recording the mass spectra of the compounds. Electronic spectra recorded in ethanol on Shimadzu UV and visible 310 IPC Spectrophotometer. Elemental analysis performed in the microanalytical center of Cairo University, Cairo, Egypt. Biological activity carried out in Biology center, Faculty of Science, Ain Shams University, Egypt.

**N-(2-Oxindolin-3-ylidene)isonicotinohydrazonic acid(1)**

Equimolar mixture of isatin and isonicotinic acid hydrazide in MeOH (100 ml) with drops of AcOH, heated under reflux for 1 h, cooled. The solid obtained filtered off and crystallized from MeOH to give 1 as orange crystals. Yield 78%, M.p: 289˚C - 290˚C. IR (\(\nu\)) cm\(^{-1}\): 3400 - 3300 (b, OH), 3190 (NH), 1697 (C=O), 1620, 1604 (C=N), 1300 (N -N), 862.6, 811 (aromatic CH). \(^1\)H-NMR (DMSO-\(d_6\)) \(\delta\) ppm: 13.99 (s, 1H, NH), 10.81 (s, 1H, OH), 8.88 - 8.81, 8.00 - 8.04 (d, d, 2H, of pyridine), 7.91 - 7.6 (m, 2H, pyridine), 7.44 - 7.07 (m, 4H, aromatic protons). \(^{13}\)CNMR (DMSO-\(d_6\)) \(\delta\) ppm: 162 (C=O \(\leftrightarrow\) C-\(\text{OH}\)), 142 (C=N), 132.21 - 121.20 (aromatic carbons), 119.54, 111.32 (C-N). Anal. Calcd: C, 63.5; H, 3.75; N, 21.05% for C\(_{14}\)H\(_{10}\)N\(_4\)O\(_2\) (266). Found: C, 63.21; H, 7.59; N, 20.85%.

**3-(Pyridin-4'-yl)-5H-1,2,4-triazino[5,6-b]indole (2)**

A mixture of 1 (5 g) and ammonium acetate (5 g) in glacial acetic acid (10 ml) heated under reflux for 2 h, cooled then washed with cold water and crystallized from EtOH to give 2. Yield 65%. M.p: 296˚C - 297˚C. IR (\(\nu\)) cm\(^{-1}\): 3191 (NH), 1624, 1619, 1597 (C=N), 1341 (NCN), 893, 846 (aromatic CH). \(^1\)HNMR (DMSO-\(d_6\)) \(\delta\) ppm: 13.98 (s, 1H, NH), 8.8 - 8.86 (d, d, 2H, pyridine), 7.78 - 7.03 (m, 4H, aromatic protons). \(^{13}\)CNMR (DMSO-\(d_6\)) \(\delta\) ppm: 149.9, 142.72, 139.2 (C=N), 132.20 (C-N), 122.84 - 121.20 (aromatic carbons), 119.54, 111.32 (C5 - C6 of 1,2,4-triazine). Anal. Calcd: C, 68.01; H, 3.64; N, 28.34% for C\(_{14}\)H\(_9\)N\(_5\) (247). Found: C, 67.65; H, 3.55; N, 28.11%.

**3-(Pyridin-4'-yl)-5-formyl-1,2,4-triazino[5,6-b]indole (3)**

A mixture of 2 (0.01 mol) and formyl acetate (0.01 mol prepared by reflux of 10 ml Ac\(_2\)O with 10 ml HCOOH for 10 min) in diethyl ether (100 ml) then stir along 2 h. The solid produced filtered off and crystallized from dioxane to give 3. Yield 60%, M.p: 304˚C - 305˚C. IR (\(\nu\)) cm\(^{-1}\): 3030 (aromatic CH), 1738 (C=O), 1589 (C=N), 880, 820 (aromatic CH). \(^1\)HNMR (DMSO-\(d_6\)) \(\delta\) ppm: 9.2 (s, 1H,
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CHO), 8.88, 8.82 (d, d, 2H, pyridine), 8.01 - 7.80 (2H, pyridine), 7.61 - 6.92 (m, 4H, aromatic protons); $^{13}$CNMR (DMSO-$d_6$) δ ppm: 164.49 (C=O), 150.86, 150.26, 144.20, 142.73 (C=N), 140.10, 139.21 (C-N), 127.11 - 121.20 (aromatic carbons); 119.55, 115.46 (C5-C6 of 1,2,4-triazine). Anal. Calcd: C, 65.45; H, 3.27; N, 25.45% for C$_{15}$H$_9$N$_5$O (275). Found: C, 65.15; H, 3.12; N, 25.30%.

**Schiff’s base (4a-c)**

A mixture of 3 (0.01 mol) and primary aromatic amines, such as 4-fluoroaniline, 3,4-dichloroaniline and/or 3-bromoaniline (0.01 mol) were heated in in EtOH (100 ml) under reflux for 1 h, cooled. The resulted solid filtered off and crystallized from EtOH to produce compounds 4a-c.

4a: Yield 60%, M.p: 297˚C - 298˚C. IR (ν) cm$^{-1}$: 2941, 2864 (aliphatic CH), 1629 (C=N), 1364 (NCN), 1255 (C-F), 912, 865, 802 (aromatic CH), 765 (C-F). $^1$HNMR (DMSO-$d_6$) δ ppm: 8.95 (s, 1H, CH=N), 8.88 - 8.87 (d, d, 2H, pyridine), 7.81 - 7.62 ( m, 2H, pyridine). $^{13}$CNMR (DMSO-$d_6$) δ ppm: 150.8 (CH=N), 142.74, 139.23 (C=N), 132.21 (C-N), 122.86 - 119.56 (aromatic C-C). Anal. Calcd: C, 68.47; H, 3.53; N, 22.8; F, 5.16% for C$_{21}$H$_{13}$N$_6$F (368). Found: C, 68.31; H, 3.25; N, 22.55; F, 5.01%.

4b: Yield 72%, M.p: 293˚C - 295˚C. IR (ν) cm$^{-1}$: 3050 (aromatic CH), 2917, 2880(aliphatic CH), 1625 (C=N), 1479 (aliphatic CH), 1364 (NCN), 910, 864 (aromatic CH), 1254, 764 (C-Cl). Anal. Calcd: C, 60.14; H, 2.86; N, 20.04; Cl$_2$, 16.94% for C$_{21}$H$_{12}$N$_6$Cl$_2$ (419). Found: C, 59.85; H, 2.69; N, 19.84; F, 16.79%.

4c: Yield 55%, M.p: 182˚C - 185˚C. IR (ν) cm$^{-1}$: 3050 (aromatic CH), 2980 (aliphatic CH), 1625 (C=N), 1488 (aliphatic CH), 1364 (NCN), 910, 864, 804 (aromatic CH), 700 (C-Br). Anal. Calcd: C, 58.74; H, 3.03; N, 19.58; Br, 18.6% for C$_{21}$H$_{13}$N$_6$Br (429). Found: C, 58.59; H, 2.89; N, 19.40; Br, 18.49%.

**Asymmetric thioethers (5a-c)**

Compounds 4a-c (0.5 g) and thiophenol (3 ml) in ethyl benzene (100 ml) heated under reflux for 4 - 6 h, cooled then added petroleum ether (40˚C - 60˚C, 100 ml). The solid thus obtained filtered off and crystallized from dioxane to give 5a-c respectively.

5a: Yield 62%, M.p: 78˚C - 80˚C. IR (ν) cm$^{-1}$: 3149 (aromatic CH), 2920, 2880 (aliphatic CH), 1618 (C=N), 1480 (aliphatic CH), 1234 (C-F), 1099 (C-S-C), 890, 867 (aromatic CH), 788 (C-F). $^1$HNMR (DMSO-$d_6$) δ ppm: 11.32 (s, 1H, NH), 7.62 - 7.60 (s, 1H, CH-N), 7.405, 7.43 (d, d, 2H, pyridine), 7.417, 7.44 (d,d, 2H, adjacent C-F), 7.29, 7.27 (m, 2H, pyridine), 7.226 - 7.208, 7.177 - 6.984 (each m, 6H, aromatic protons). $^{13}$CNMR (DMSO-$d_6$) δ ppm: 175 (C -S), 162.61 (S -C-N), 143.71 (C-F), 132.13 (C=N), 128.25 - 122.71 (aromatic carbons), 119.11, 111.38 (C5 - C6 of 1,2,4-triazine). Anal. Calcd: C, 67.78; H, 3.97; N, 17.57; F, 3.97; S, 6.69% for C$_{27}$H$_{19}$N$_6$FS (478). Found: C, 67.59; H, 3.85; N, 17.33; F, 3.79; S, 6.55%.

5b: Yield 75%, M.p: 59˚C - 60˚C. IR (ν) cm$^{-1}$: 3085 (aromatic CH), 2950, 2885 (aliphatic CH), 1710 (C=O), 1610 (C=N), 1488 (deformation CH$_3$), 1080 (C-S-C), 910, 880, 810 (aromatic CH), 700 (C-Cl). Anal. Calcd. C, 61.24; H, 3.40; N, 15.90; Cl, 13.42; S, 6.04% for C$_{27}$H$_{19}$N$_6$Cl$_2$S (529). Found C, 61.18; H, 3.25; N, 15.81; S, 5.89; Cl, 13.38%.
5c: Yield 58%, M.p: 68°C - 70°C. IR (ν) cm⁻¹: 3065 (aromatic CH), 2910, 2880 (aliphatic CH), 1700 (C=O), 1610 (C=N), 1488, 1444 (deformation of CH₃), 1090 (C-S-C), 915, 880, 820 (aromatic CH), 680 (C-Br). Anal. Calcd: C, 60.11; H, 3.52; N, 15.58; Br, 14.84; S, 5.93% for C₂₇H₁₉N₆BrS (539). Found: C, 59.89; H, 3.45; N, 15.39; Br, 14.61; S, 5.80%.

- **3-Aryl-5-methyl-2-(3-(pyridin-4-yl))-5H-[1,2,4]triazino[5,6-b]indol-5-yl) thiazolidin-4-ones (6a-c)**

A mixture of 4 (0.01 mol), thiolactic acid (0.05 mol) and sodium acetate anhydrous (5 g) in dry dioxane (100 ml) was heated under reflux for 6 h, cooled then poured onto ice and neutralized with NaHCO₃. The solid produced filtered off and crystallized from EtOH to give compounds 6a-c.

**6a:** Yield 66%, M.p: 210°C - 212°C. IR (ν) cm⁻¹: 3163 (aromatic CH), 2900, 2880 (aliphatic CH), 1718 (C=O), 1616 (C=N), 1463 (aliphatic CH), 1234 (C-F), 1149 (C-S-C), 983, 846 (aromatic CH), 746 (C-F). ¹HNMR (DMSO-d₆) δ ppm: 11.43 (s, 1H, OH of ( ), 8.95 (s, 1H), 8.88, 8.87 (d,d, 2H, pyridine), 7.81, 7.80 (each d, d, 2H, C-F), 7.79 - 7.52 (m, 2H, of aryl), 7.42 - 7.40 (2H, pyridine), 7.39 - 6.98 (m, 4H of benzo protons), 3.37 (s, 1H, C5-αH, thiazolidinone), 2.51 (s, 3H, Me); ¹³CNMR (DMSO-d₆) δ ppm: 163.1 (C=O), 150.85 (C-S), 142.75 (C-F), 139.23,135.78 (C=N), 129.45 - 121.21 (aromatic carbons), 119.57, 111.33 (C5 - C6 of 1,2,4-triazine), 39.52 (CH₃).

Anal. Calcd. C, 60.26; H, 3.27; N, 18.34; F, 6.98% for C₂₃H₁₅N₆FSO₂ (458). Found: C, 60.11; H, 3.02; N, 18.12; F, 4.35; S, 6.80%.

**6b:** Yield 70%, M.p: 228°C - 230°C. IR (ν) cm⁻¹: 3080 (aromatic CH), 2950, 2880 (aliphatic CH), 1610 (C=N), 1488, 1444 (deformation of CH), 1085 (C-S-C), 910, 890, 815 (aromatic CH), 710, 700 (C-Cl). Anal. Calcd: C, 54.22; H, 2.75; N, 16.50; Cl₂, 13.94; S, 6.28% for C₂₃H₁₄N₆Cl₂SO₂ (509). Found: C, 53.95; H, 2.55; N, 16.18; Cl₂, 13.85; S, 6.01%.

**6c:** Yield 58%, M.p: 215°C - 218°C. IR (ν) cm⁻¹: 3055 (aromatic CH), 2900 (aliphatic CH), 1610 (C=N), 1580 (C=N), 1480 (deformation CH), 1090 (C-S-C), 910, 870, 820 (aromatic CH), 700 (C-Br). Anal. Calcd: C, 53.17; H, 2.89; N, 16.18; Br, 15.41; S, 6.16% for C₂₃H₁₅N₆BrSO₂ (519). Found: C, 52.89; H, 2.71; N, 15.95; Br, 15.19; S, 5.89%.

**6. Conclusion**

A simple route to obtain various halogenated 4-thiazolidinones bearing a 1,2,4-triazino[5,6-b]indole moiety as fungicidal agents have been deduced from condensation of 5-formyl-1,2,4-triazinoindole with halogenated primary aromatic amines, followed by 1,2-addition of thiophenol and/or thiolactic acid. The fluorinated substituted showed a higher activity towards the tested fungi.

**Conflicts of Interest**

The author declares no conflicts of interest regarding the publication of this paper.
References


