Evaluation the Vickers Hardness of Denture Base Materials

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Abstract

The aim of this study was to evaluate the Vickers Hardness of four different denture base materials. Four different denture base materials were used in this study: Heat-cured pink acrylic resin, self-cured acrylic resin, heat-cured clear acrylic resin and polyamide denture base materials. Totally 80 disc shaped samples which 2 mm thickness and 40 mm diameter were prepared. The samples were stored in distilled water after polished. Then, Vickers Hardness test was performed with TMTech HV-1000B (200 gram at 15 second). The statistically analysis of obtained data was performed with use one-way analysis of variance test. The results showed that the highest Vickers Hardness value (18.57 HV) was identified in heat-cured (pink) acrylic resin, the lowest Vickers Hardness value (9.62 HV) was identified in polyamide denture base materials. The result of one-way analysis of variance test showed that there were statistically very highly significant differences between four denture base materials of Vickers Hardness (p < 0.001). The obtained data presented that the self-cured denture base acrylic resin was the hardnest denture base materials.

Keywords

Hardness, Denture Bases, Polyamide

1. Introduction

In the clinical practice of dentistry, there are many circumstances in which the need to adjust denture base acrylic resin is necessary [1] [2].

In 1935, first acrylic resin was produced by the imperial kolladent chemical industry.

In 1937, heat-polymerized acrylics which developed by Walter H. Writer, have been used extensively as den-
ture base materials. Acrylic polymers has been successfully used in various areas such as shutters, denture base
materials, artificial teeth, denture repair materials, crowns and bridges face the vestibular prosthesis [3] [4].

Denture base polymers are usually supplied as a mixture of poly methyl methacrylate (PMMA) powder beads
and methyl methacrylate (MMA) monomer liquid [5]. MMA: Acrylic resin most commonly used in dentistry is
metil methacrylate which is the methyl ester of methacrylic acid. The polymethyl methacrylate is formed by the
polymerization of styrene. Since 1946, 98% of all denture base polymers MMA and copolymers began to be
performed. Although polymerization could be made ultraviolet and visible light; chemical initiator is commonly
used for polymerization in dentistry. During the polymerization of pure methyl methacrylate up to 21% ratio of
a volume shrinkage occurs. PMMA: pure form is a transparent resin. The 0.25 micrometer wavelength UV light
can pass even through the acrylic and not be discolored by UV. to create similar colors with tissue, can be used
together with many pigment. The advantages of PMMA include excellent esthetic properties, adequate strength,
low water sorption, lack of toxicity, facility of repair, and construction by a simple molding and processing
technique [6].

The chemically polymerizing acrylics were used for dental purposes first in Germany during World War II. in
order to separate from heat polymerized acrylic resin, the “self polymerizing” or “cold polymerized” or “auto
polymerized” is called. These are generally used as a repair acrylic, as well can be used as a base acrylic. Com-
pared with heat polymerized resins; the amount of residual monomer and water absorption is higher in the che-
merically polymerized resins, dimensional stability is less. Since it includes amin accelerators, it is worse com-
pared to polymerization of resins with color harmony. The particle structure is more irregular. Also; the struc-
tures weaker and more flexible for the polymer particles have a lower molecular weight [7].

Alternative polymer systems to PMMA, such as polyamide, epoxy, polystyrene, or vinyl acrylic resins, have
also been tried. However, the desired denture base material has not been developed yet [8] [9].

A removable partial denture without metal clasps has recently been used in dental practice. In recent years,
injection-molded thermoplastic resins such as polycarbonate, polyamide, and polyester have been used as den-
ture-base materials [10] [11]. Injection molded thermoplastic resins (polymides, polyethylene terephthalate,
and polycarbonate) are used for denture bases of RPDs without metal clasps because of their advantageous char-
acteristics, such as a higher elasticity than heat polymerizing base resins, and the fact that they can facilitate
denture retention by utilizing the undercuts of abutment teeth in the denture base design [12]. Polyamides,
known as “nylon” are thermoplastic polymers produced by condensation between a diamine and a dibasic acid
[13].

Firstly, PA was used as a denture base polymer in the studies in 1950 s [14] and reported that because of the
high water absorption and discoloration, the nylon was not suitable materials in dentistry [15].

The aim of this study was to evaluate the Vickers Hardness of four different denture base materials and inves-
tigate to the availability of dental prosthesis. The hypothesis this study was that the heat-cured (pink) acrylic re-
sin will has the highest Vickers Hardness value.

2. Materials and Methods

Four different denture base materials were used in this study: Heat cured pink acrylic resin QC 20 (De Trey,
Dentsply, Addlestone, UK), self-cured acrylic resin Takilon (cold-curing dental polymer, Rodont, Italy), heat
cured clear acrylic resin Vertex (Vertex Dental B.V., Zeist, Netherlands) and polyamide denture base materials
Deflex (Nuxen S.R.L., Buenos Aires, Argentina) (Figure 1). Totally 80 disc shaped which 2 mm thickness and
40 mm diameter were prepared. Firstly wax models were prepared like the size of specimens. They were em-
bedded to dental stone in the stainless steel mold and the mould was created for the acrylic resin. Each acrylic
material was mixed according to the manufacturer’s instructions and placed to the mould. All specimens were
pressed under the 3000 PSI with hydraulic presser. After the acrylic resins were polymerized heat and self.
The polyamide dentures were injected with 720 - 750 kPa pressure at 220˚C and pre-heating time of 15 min accord-
ing to manufacturer’s instructions. After the specimens were prepared, all irregularities were removed with a tungsten carbide bur. The samples were polished and stored in distilled water at 4˚C until use.

The Vickers Hardness test was performed with TMTeck HV-1000B (TMTeck Manufacturing Limited, Bei-
jing, China) (200 gram at 15 second) to all specimens (Figure 2). The strength was applied to midpoint of base
materials by diamond tip. Then, diagonals of trace whose shape is square was measured by microscope and
evaluated their values of hardness. The statistically analysis of obtained data were performed with use one-way
analysis of variance test by IBM SPSS Statistics 16.
3. Results

The result of one-way analysis of variance test showed that there were statistically very highly significant differences between four denture base materials of Vickers Hardness (p < 0.001) (Table 1).

The results showed that the highest Vickers Hardness value (18.57 HV) was identified in heat-cured (pink) acrylic resin, the lowest Vickers Hardness value (9.62 HV) was identified in polyamide denture base materials (Table 2, Figure 3).

4. Discussion

The hypothesis of this study was accepted, heat-cured (pink) acrylic resin has the highest Vickers Hardness value.

The acrylic resins are usually used in the removable prosthesis for the denture base materials. Acrylics are hard materials so they may disturb the patient during use. Despite the use of soft lining materials, they can cause the canker sores and soft lining materials must be changed over time. The polyamide dentures are used as an alternative for acrylics in removable prosthesis. The polyamide dentures are lighter and have a flexible structure.

Because of the easy specimen preparation, simplicity of the test method and availability of the equipment, hardness has been widely used as a method of investigating factors that influence the degree of conversion of resins and for characterization of the mechanical quality of a polymer [6] [16]-[18]. In this study, Vickers Hardness test was performed to the samples.
### Table 1. The one-way analysis of variance.

<table>
<thead>
<tr>
<th>Source</th>
<th>Sum of Squares</th>
<th>df</th>
<th>Mean Square</th>
<th>F</th>
<th>Sig.</th>
</tr>
</thead>
<tbody>
<tr>
<td>Between Groups</td>
<td>862.120</td>
<td>3</td>
<td>287.373</td>
<td>114.074</td>
<td>0.000</td>
</tr>
<tr>
<td>Within Groups</td>
<td>191.459</td>
<td>76</td>
<td>2.519</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Total</td>
<td>1053.579</td>
<td>79</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

### Table 2. Mean HV value of denture base materials (N = 20) (X = HV).

<table>
<thead>
<tr>
<th>Material</th>
<th>Mean</th>
<th>Std. Deviation</th>
<th>Std. Error</th>
<th>95% Confidence Interval for Mean</th>
<th>Minimum</th>
<th>Maximum</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Mean</td>
<td></td>
<td></td>
<td>Lower Bound</td>
<td>Upper Bound</td>
<td></td>
</tr>
<tr>
<td>Heat-Cure (Clear)</td>
<td>16.23</td>
<td>1.85202</td>
<td>0.41412</td>
<td>15.3622</td>
<td>17.0958</td>
<td>12.32</td>
</tr>
<tr>
<td>Heat-Cure (Pink)</td>
<td>18.57</td>
<td>1.35209</td>
<td>0.30234</td>
<td>17.9372</td>
<td>19.2028</td>
<td>16.91</td>
</tr>
<tr>
<td>Self-Cure</td>
<td>14.70</td>
<td>1.64694</td>
<td>0.36827</td>
<td>13.9292</td>
<td>15.4708</td>
<td>10.61</td>
</tr>
<tr>
<td>Polyamide</td>
<td>9.62</td>
<td>1.45130</td>
<td>0.32452</td>
<td>8.9398</td>
<td>10.2982</td>
<td>8.26</td>
</tr>
<tr>
<td>Total</td>
<td>14.78</td>
<td>3.65191</td>
<td>0.40830</td>
<td>13.9668</td>
<td>15.5922</td>
<td>8.26</td>
</tr>
</tbody>
</table>

### Figure 3. The Vickers Hardness graph of denture base materials.

During the polymerization reaction, the conversion of monomer into polymer is not complete and varying amounts of free or unreacted monomer remain in the polymerized resin [17] [19]-[24]. Residual monomer is a well-known plasticizer and affects the mechanical properties of the acrylic resins [17] [18] [25] [26]. Residual monomers can be polymerized again in the patient’s mouth and damage to the patient’s mucosa and crest. Considering that self-curing acrylic resins usually exhibit high residual monomer content [17] [23] [24]. For this reason, we usually prefer to heat curing acrylic resins at dental prosthesis. On the other hand, self-curing acrylic resins often use repairing of broken dentures, making the night-guard.

This finding was not to support the other studies, which have shown that the hardness of self-curing acrylic
resins is lower than that of heat-cured acrylic resins [17] [25] [27]. The values found in this study showed that, the self-curing acrylic resins have higher Vickers Hardness value from heat-curing acrylic resins. The reason of this, acrylic can be turned into a fragile structure at high temperatures. Braun et al. [25] have found that, when submitted to chemical polishing, the heat-cured material was harder than the self-cured one. Another study, Ayaz et al. [28] reported that heat and microwave polymerization techniques did not affect the hardness property significantly.

Heat activates the chemical reaction between the monomer and polymer components of the resin and produces more complete polymerization [29]. This mechanism may explain why hot water conditions improved the mechanical properties of resin.

Pinto et al. [30] reported that polyamide resins had a higher mechanical resistance than acrylic resins. In this study, polyamide had the lowest Vickers Hardness values.

5. Conclusion

Within the limitations of this study, it can be concluded that, the head-curing (pink) acrylic resins had the highest Vickers Hardness and polyamide was not as hard as the other materials. The polyamide dentures more should be used in the removable prosthesis to minimize the trauma of prosthesis and the plants of patients.

Acknowledgements

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References


