

# Literature Review on Biorefinery Processes Integrated to the Pulp Industry

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# Abstract

Biorefinery is a new term to designate two main subjects, bioproducts and bioenergy, which play important roles towards a more bio-based society. This paper reviews the current biorefineries model as well as its future importance focusing on pulp mill opportunities. There are currently several different levels of integration in biorefineries which add to their sustainability, both economically and environmentally. Enzymatic pretreatment for biomass deconstruction aiming to release the polysaccharides is a key technology in the future biorefineries and it is currently the subject of intensive research.

# Keywords

Lignocellulosic Biomass, Biorefinery, Ethanol

# **1. Introduction**

A great challenge for the modern civilization is becoming a sustainable society based on more realistic needs and rational utilization of the natural resources. In this context, energy is one of the most important issues. Currently, the fossil fuels (oil, coal, and natural gas) are the main energy sources worldwide but their exploitation is neither sustainable nor environmental friendly.

The scientific community believes that in the near future large volumes of biomass will be processed to produce biofuels together with valuable co-products, which will be increasingly used to make other bioproducts [1]-[6]. In this scenario, certain bio-based industries such as the pulp and the sugar cane could take enormous advantage of their current knowhow in order to make intelligent use of their biomass feedstocks, by operating their facilities as biorefineries. Biorefinery is a new term that is related to two main subjects, value-added bio-

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products (chemical building blocks, materials, pulps), and bioenergy (biofuels, power, and heat) from biomass, considering sustainability assessment and life cycle [7].

Many recent studies have pointed out the pulp industry as an interesting platform to deploy the biorefinery concept [8]-[10]. This industry applies partial biomass deconstruction processes for producing pulp and other products, such as paper, viscose, acetate, nitrates, lignosulphonates, vanillin, ethanol, acetic acid, etc., in addition to steam and power. The processes currently used are efficient to separate the lignin and extractives fractions from the polysaccharides; this cracking is very suitable and necessary for biorefinery applications. This paper reviews the current biorefineries model as well as its future importance focusing on cellulosic pulp mill opportunities.

# 2. Biorefineries

The emerging bio-based economy is a very promising sector with notable future potential and many business opportunities [11]; in this scenario it is includes the biorefinery concept. There are many different definitions for biorefinery, but in general the term is applied for using renewable raw materials (e.g. biomass) to produce energy together with a wide range of everyday commodities in an economic and sustainable manner [3] [7] [12] [13]. The fossil fuels (oil, coal, and natural gas) are the main energy source worldwide; however, one of the most important challenges of the modern society is in decreasing our dependency on fossil fuel reserves and boosting rural development. The biorefinery concept attempts to apply to biomass conversion, the methods that have been applied to the refining of petroleum [3]. Biorefineries would simultaneously produce biofuels as well as biobased chemicals, heat, and power. Biorefineries are seen as a very promising route to meeting our aims for sustained prosperity and preserving the environment. Unlike oil refineries, which are units of large industrial size, biorefineries should include facilities of various sizes [14].

Renewable sources of energy are required for sustainable development of our society in the near future [15]. Plant biomass is the main source of renewable materials on earth and represents a potential source of renewable energy. However, the main source of bioethanol currently used (starch from cereal seeds) is too expensive and of limited availability. By contrast, cellulose is available in high amounts at very low cost (as forest, agricultural or industrial lignocellulosic wastes and cultures) and, after overcoming some current barriers, could be a widely available and inexpensive bioethanol source in the future [7] [16] [17].

# **3. Biomass Sources**

Hardwoods and softwoods have been the world's main raw materials for pulp production and they deliver suitable product quality. However, the cost of these raw materials has greatly increased in the last decade due to many factors, including the newly created demand for biorefinery applications. Therefore, the establishment of new raw materials that can compensate the lack of low cost wood has been largely sought. In this way, each one should be able to identify the ideal biomass source which should be used for biorefinery application. High productivity plants such as grasses can potentially supply biomass of low cost to meet the current demand [18]-[22]. Currently, the biomass sources are divided in two groups: woody and nonwoody.

Forest and agricultural residues have been the main target in Europe, North America and other places of temperate climate. In tropical climates like in Brazil there is great opportunity to use rapid growth and high productivity woody and nonwoody biomass to practice the Biorefinery concept. In fact, it is possible harvesting eucalypt plantations at one year of age, resembling agricultural crops, with fairly high biomass production. This has been possible due to the great improvements achieved in eucalypt biomass productivity through intensive breeding programs in the last 50 years. There exist clonal plantations in Brazil that delivers productivities of 40 ton/ha/year of bone dry wood at the age of 7 years [23]. Planted forests are very productive and homogeneous, delivering outstanding wood quality. Another relevant point is that these breeding programs have been focused on pulp and paper production and the wood quality has been tailor-made for that. When the focus is Biorefinery, there is much more to be done in regard to the understanding of the wood chemical characteristics required for this new purpose.

Regarding woody raw materials some countries and species have been highlighted, e.g. in Brazil has a forest vocation spurred by its large territorial area and tropical climate. The forested area covers more than half of the Brazilian territory (forest per capita: 2.56 ha/person), but only 7 million ha (0.7%) is made of planted forests [24]. The Brazilian planted forests (6.7 million ha) are dominated by euclypts (over 5.1 million ha) and *Pinus* 

spp (over 1.6 million ha). Some of the Brazilian planted forests are highly productive, particularly those of *Eucalyptus* spp (**Table 1**). In some areas of the Brazilian southeast and northeast *Eucalyptus* spp productivity can reach 80 m<sup>3</sup>/ha/year [25]. The very high productivity of Brazilian eucalypt forests stems from the good adaptation of the species to the climate but also from the breakthrough scientific and technological advances made by Brazilian eucalypt forests increased from 24 m<sup>3</sup>/ha/yr in 1980 to 41 m<sup>3</sup>/ha/yr in 2012, representing an 83% increase in productivity of planted forests in Brazil. It is expected that this productivity will reach 70 m<sup>3</sup>/ha/year in the near future [23].

Regarding nonwoody raw materials, there are many possibilities. Sugar cane is one of the most important biomass sources in the world, since it can be used as a food resource (e.g. sugar) or raw material to produce renewable fuels (e.g. ethanol). The sugar cane (*Saccharum* spp) is an interesting raw material for biorefinery since it can provide the following materials: 1) sugars (sucrose, glucose, and xylose); 2) natural polymers (cellulose, hemicellulose, and lignin present in the bagasse and straw); 3) stillage rich in organic raw material; and 4) other compounds, such as alcohols and carbon dioxide ( $CO_2$ ).

Currently, the biorefinery concept applied to sugar cane has focused on the production of first generation ethanol (from extracted sugar cane juice), and second generation ethanol (from cellulose). There are three new plants planned for the year 2014 for producing second generation ethanol in Brazil using a biorefinery model [24]. The second generation ethanol will largely enhance the total ethanol that can actually be produced from sugar cane. For example, the average sugar cane productivity in Brazil is 85 tons per hectare, and for each ton of sugar cane processed about 140 kg dry-matter of bagasse and 140 kg dry-matter of straw are generated, *i.e.*, 12 tons of straw and 12 tons of bagasse per hectare [5]. The estimated current total global production of first generation ethanol from sugar cane is 20 billion liters a year [25]. When using the sugar cane bagasse and straw in a biorefinery model, the theoretical ethanol yield (considering that all glucose will be converted to ethanol) would be improved by approximately 98.4%, *i.e.*, the current production could be increased to 49.7 billion liters a year [5].

The use of the sugar cane bagasse residue after sugar extraction to produce second-generation biofuel (e.g. ethanol) has been extensively studied due to its wide availability and low cost [5] [26]. However, the process of using sugarcane bagasse residues to produce ethanol is not cost-effective due to high energy consumption required for a harsh pretreatment process, a complicated detoxification process and the use of expensive enzymes for polysaccharide hydrolysis [27].

Another potential nonwoody raw material is the elephant grass (*Pennisetum* spp) which is highly effective in the CO<sub>2</sub> fixation during the photosynthesis process for biomass production. Reference [2] [18]-[20], obtained different results of biomass production from elephant grass (cultivar "Paraíso") due to the various treatments in their research works. The total production of biomass with 85% of dry matter is 27 ton/ha and may be stored for an indefinite period of time. In accordance with reference [22], the advantages of elephant grass as biomass source in relation to other sources are: greater productivity (45 t dry-matter/ha/year), shorter productive cycle (two cuts per year), better cash flow, possibility of total mechanization, and greater assimilation of carbon. Some characteristics of elephant grass compared with sugar cane bagasse are reported by reference [28]. Elephant grass has special potential for biorefinery applications because of its high fiber production of adequate chemical composition, similar to sugar cane bagasse in many senses [28] [29]. Some studies on elephant grass have shown a

Table 1. Annual productivity of main Brazilian planted forests [24].					
Species	Productivity, m³/ha/year				
Acacia (Acacia spp)	15 - 25				
Araucaria (Araucaria angustifolia)	10 - 25				
Aspen (Populus deltoides)	20 - 30				
Eucalypt (Eucalyptus spp)	30 - 40				
Pine (Pinus spp)	25 - 30				
Teak (Tectona grandis)	15 - 20				

content of 40%, 30% and 17.7% for cellulose, hemicelluloses and lignin, respectively [30]. These values are adequate for biorefinery applications, in special the low lignin content that indicate an easier fractionation of this material in deconstruction processes.

Another characteristic of elephant grass is that in the stalks, the fiber content is frequently higher than in leaves (53% and 45% - 46%, respectively), as well as lignin (11% and 9%, respectively) and cellulose (40% - 41% and 29% - 30%, respectively) [29]. Another important parameter to take into account is the amount of energy produced per hectare per year. In average, elephant grass can produce 190,000 Mcal/ha/year whereas eucalypt produces about 100,000 and sugar cane bagasse only 30,000.

## 4. The Pulp Mills as Biorefineries Platform

The major components of all lignocellulosic materials are cellulose, hemicelluloses, and lignin, which have the structural formulae  $CH_{1.67}O_{0.83}$ ,  $CH_{1.64}O_{0.78}$ , and  $C_{10}H_{11}O_{3.5}$ , respectively [31]. There are also other compounds in minor proportion such as proteins, lipids, pectin, soluble sugars and minerals [32] [33]. In its native state, lingo-cellulosic material is recalcitrant to efficient hydrolysis of its cellulose fraction into glucose monomer due to the physicochemical and structural composition of the material [34]. The most important carbohydrates present in the biomass plant are shown in the **Figure 1**. Partial deconstruction of biomass to make paper pulp is a quite well known and established technology and aimed to separate cellulose and hemicelluloses fractions from the lignin, extractives and minerals. However, deconstruction aimed at production of biofuels and biomaterials other than paper pulp is still a great challenge [35] [36]. Although lignocellulosic biomass is a very promising biofuel and bioproducts source, it cannot be currently commercially exploited mainly due to limitations for chemical or enzymatic hydrolysis caused by the presence of lignin. If this drawback is solved, low cost wood materials can be the source of choice to obtain renewable energy in the near future.

Thus, the optimization of pretreatment methods for deconstructing biomass into their cellulose, hemicelluloses and lignin fractions is of great importance for the pulp and paper, biofuel and bioproducts industry [37]-[39]. Such optimization requires a particular emphasis on the development of analytical methods for the characterization of the structural (cellulose, hemicelluloses, lignin) and non-structural (lipids, minerals, etc.) biomass constituents and the study of their fate during the pretreatment and deconstruction process. This is particularly true when dealing with unknown faster growing feedstocks such as elephant grass.

Partial deconstruction of lignocellulosic biomass to make paper pulp is currently done with the secular kraft process, which delivers a high quality product. This process is efficient to separate the lignin and extractives fractions from the polysaccharides, but the lignin fraction becomes contaminated with sulfur, which hinders its utilization in many applications and creates air pollution problems. If lignin were free from sulfur, they could potentially be more easily recovered for producing energy and as a phenol source for many applications such as carbon fiber, activated carbon and other aromatic added-value chemicals [40] [41]. Among the non-sulfur partial deconstructing biomass processes currently known, the soda-anthraquinone (AQ) and organosolv processes are included [42]. Partial deconstruction by these methods have been largely studied in the past and proven non-





competitive with the kraft process mainly due to the low paper pulp product quality.

Nevertheless, in the context of operating a pulp mill as a Biorefinery, where wood is not the unique feedstock and paper pulp is not the only product within the value added product chain, such processes may become more interesting than the kraft (Figure 2). For example, the quality of the paper pulps derived from these processes may be improved by incorporating to them some fractions of the deconstruction process such as hemicelluloses [43]. Other paper products less demanding in regard to strength can be made from these pulps without penalty [43]. A significant part of the pulp produced can be actually directed towards biofuel and biomaterials production, and in this case strength is not an issue. Furthermore, in all these alternative pulping methods the black liquor derived from biomass deconstruction is much more amenable for recovery than that of the kraft process, both for energy and lignin separation and conversion into biomaterials.

# 5. The Biochemical Platform in Lignocellulose Biofuels

Another option that a biorefinery should contemplate is the complete biomass deconstruction without going through paper pulp production. This can be done by two different routes, the biochemical and the thermochemical conversions. In both cases biochemical intermediates and residues are produced with the first being used for fuels, chemicals and materials production, and the residues used for combined heat and power generation. Although both routes are promising, the biochemical platform integrates better with a paper pulp manufacturing facility.

The biochemical platform converts the carbohydrate fraction into fermentable sugars, leaving lignin and other fractions as a residue. There are several approaches to this, including: 1) separate hydrolysis and fermentation; 2) simultaneous saccharification and fermentation; 3) simultaneous saccharification and co-fermentation and 4) consolidated bioprocessing [5] [44]. In the first approach the C5 sugars are firstly hydrolyzed under acidic conditions and the solids remaining are treated with cellulases to produce C6 fermentable sugars, leaving lignin and other materials as residues. In the second case the C5 sugars are firstly hydrolyzed under acidic conditions and the solids remaining are scarified/fermented simultaneously to produce bioethanol, leaving lignin and other materials as residues. In both cases, the C5 and C6 sugars are fermented separately. The third scenery suggests a similar route as the second, except that C5 and C6 sugars are fermented together. The fourth route is still a theory, but may become reality in the future, and suggests the acid hydrolysis of C5 sugars followed by cellulases production, hydrolysis of C6 and co-fermentation of C5 and C6 in a single reactor, leaving lignin and other materials as residues.

In the biochemical platform, acid hydrolysis is the conventional method for the liberation of simple sugars for fermentation; however, it is not the ideal method, since compounds such as furfural and hydroxymethylfurfural are formed as side products, which are toxic for the yeasts, and thus lower the fermentation yield [5]. To avoid the formation of toxic compounds, it would be desirable to accomplish the saccharification step enzymatically using a combination of hydrolytic enzymes. However, the complex structure of the fiber wall, being composed of a matrix of entangled polymers of cellulose, hemicellulose, lignin and protein, depending on the plant material, complicates the enzymatic hydrolysis. In practice, a pretreatment stage is required to loosen the cell wall struc-



Figure 2. Pulp mill working as a biorefinery platform for producing chemicals, energy, fuels, and materials.

ture and/or remove lignin and other non-carbohydrate components in order to expose cellulose and the hemicelluloses [37] [38] [45]-[47].

## 5.1. Pre-Treatments Based on the Bost Usual Pulping Technologies

Pretreatment refers to the mechanical, physical, chemical, and/or biological treatments to reduce the particle size of the material and disrupt its cell structure to make it more accessible to chemical or enzymatic hydrolysis treatments [5] [48]. More specifically, the aims of pretreatment are typically the hydrolysis of hemicelluloses and reduction of crystallinity and degree of polymerization of cellulose, to facilitate the subsequent enzymatic hydrolysis of cellulose [48].

The most used pretreatments include: 1) Ball mill, which is based on a type of grinder used to grind biomass into extremely fine powder, being effective for a decrease in the cellulose crystallinity and improve the enzyme digestibility [49]; 2) Diluted acid, which is based on the use of diluted acids such as sulfuric and phosphoric acid at high temperature (140°C - 200°C) for up to 1 hour, aiming the hydrolyzes of the hemicelluloses fraction of the biomass to soluble sugars while the lignin condenses and precipitates [45] [50] [51]; 3) Ammonia Recycle Percolation—ARP, which is a pretreatment method based on aqueous ammonia, aiming the enzymatic digestibility and also achieves high degree of delignification [52] [53]; 4) Alkaline pretreatments: sodium hydroxide and cal*cium hydroxide*, they causes the lignocellulose to swell, thereby increasing the surface area while reducing the degree of polymerization and crystallinity of the material, these processes can be performed at ambient temperatures and pressures, although longer reaction times may be needed in this case [37] [45] [51] [54]; 5) Ozonolysis, which is based on biomass ozone application, this pretreatment has shown its efficacy essentially degrading the lignin polymer but also slightly solubilizing hemicellulose content of lignocellulosic biomass [55] [56]; 6) Organosolv, which is based on the organic solvents or their aqueous solution for extracting lignin from lignocellulosic raw materials at high temperatures; the main difference between organosoly pretreatment and organosolv pulping process is the delignification degree, which is not demanded to be as high as that for pulping [57]; 7) Biological, which is based on the use of microorganisms and their enzymatic machineries to break down lignin and alter lignocellulose structures [58]; 8) Steam explosion, which is based on the use of water at high temperature in a pressurized reactor prior a rapid decompression, providing an explosion, which effectively breaks bonds between the lignin and hemicellulose fractions, allowing the material to expand and expose the cellulose; 9) Ammonia fiber expansion—AFEX, is similar to that of steam explosion, except the biomass soaks in liquid ammonia instead of water in a pressurized reactor prior to rapid decompression; in this process the recycle of the ammonia would be required to make this option economically feasible [51]. In Table 2 the effects of various pretreatments are presented.

## 5.2. Kraft and Soda-AQ Pulping of Fast Growth Feedstocks

Biomass deconstructions through the alkaline process are quite well known and are the most used processes for converting the wood into pulp. Currently, the main pulping process used worldwide is the kraft one, which uses sodium hydroxide and sodium sulphide as reagents; its hegemony in the world comes from the fact that it produces pulp of high strength, uses any type of biomass, possesses a highly developed chemical recovery technology and is the most selective of all chemical processes known [25] [59] [60].

Concerning the alkaline processes, these are well established technologies for pulping, however targeting a new pulp industry working as a biorefinery, some characteristics of the processes should be considered, such as sulfur free technologies, since compounds containing this element are responsible for decreasing the black liquor heating value, for example [59]. Additionally, benefits of using sulfur free processes may be also interesting for decreasing environmental pressures, since the elimination of sulfur from the processes will greatly decrease aerial emissions of the reduced sulfur compounds usually produced by the kraft process [42]. Soda process also appears as an interesting alternative, since the mill structure is compatible with the kraft process. The interest in sulfur free pulping has centered on soda pulping with anthraquinone (AQ) additions, since its presence accelerates delignification during soda pulping and produces pulps with strength properties which approach kraft levels [42] [60].

Regarding the pulping process characteristics, selectivity is a significant parameter when the pulp strength properties are relevant. In this case the kraft process is the ideal choice, since it is well known to deliver the best fibers. Although selectivity is still important in the deconstruction for biofuel and bioproducts (yield) it play a

Pretreatment		Biomass changes				
		Cellulose	Hemicellulose	Lignin	Advantages	Disadvantages
Physical	Ball mill	High crystallinity decrease	Not changed	Not changed	Crystallinity decrease	High energy consumption
Chemical	Diluted acid	Mild depolymerization	80% - 100% removal	Mild removal	High xylans production	Hard acid recovery, corrosive, and high cost
	Sodium hydroxide	Significant swelling	High solubility, >50%	High solubility, >50%	Effective esters removal	High cost
	ARP	Low depolymerization	High solubility, >50%	High solubility, $\approx 70\%$	Effective delignification	High cost
	Calcium hydroxide	Mild depolymerization	Significant solubility	Partial solubilization (≈40%)	Effective lignin and acetyl removal; low cost	Low solubulity
	Ozonolysis	No depolymerization	Low solubilization	High solubility, >70%	Effective delignification	High cost, and requires ozone
	Organosolve	Significant swelling	80% - 100% removal	80% - 100% removal	High xylans production, and Effective delignification	High cost for solvent recovery
Biological	Biological	20% - 30% of depolymerization	High solubility, >80%	Partial solubilization (≈40%)	Low energy consumption, and effective delignification	Loss cellulose, low hydrolysis rate
Combined	Steam explosion	Mild depolymerization	80% - 100% removal	Mild removal, however the lignin structure is modified	Low energy consumption, and no recovery costs	Xylans degradation working as inhibitor compound
	AFEX	Decrease in the crystallinity degree	High solubility, >60%	Partial solubilization (10% to 20%)	Loss xylans, and no inhibitor formation	Recovery of ammonia, is not effective for high lignin concentrations

#### Table 2. Effect of different pretreatments in lignocellulosic biomass [5].

less important role since the fiber integrity is no longer important. In this way, the resulting black liquor free of sulfur offers more opportunities for lignin use than the kraft one.

Most modifications of the kraft process have been aimed at improving yield and the properties of the pulp produced. AQ has been used as an additive to the soda and kraft pulping processes due to improved delignification rate and protection of cellulose and hemicelluloses chains against peeling reactions. The benefits of using AQ in eucalypt kraft pulping have been proven by many authors [42] [60]-[62].

Emissions of total reduced sulfur compounds (TRS) in a kraft pulp mill have been an environmental concern. They are corrosive and responsible for the bad odor of the kraft process. TRS compounds are generated during pulping and are liberated mainly from digester together with steam. TRS formation during pulping depends on sulphidity level, pulping time and pulping temperature. Some decrease in TRS compounds emissions have been obtained by reducing pulping sulphidity [42] [63]-[65]. Use of low or zero sulfidity and addition of AQ can be an alternative to reduce pollutant emission without decreasing delignification efficiency [42] [66]. One alternative that can be used is the Soda-AQ process.

The use of Soda process catalyzed by AQ would have additional benefits by reducing mill odors, simplifying the recovery process, and allowing gasification technology to be utilized for recovery. On the other hand, the Soda-AQ process has well-known drawbacks, including the high AQ cost and the production of pulp with lower tear strength [67]. These drawbacks account for the limited implementation of Soda-AQ pulping to a few hard-wood pulp mills. On the other hand, if the Soda-AQ pulp is used to produce biofuel rather than paper pulp, the strength problem is not a problem. The financial considerations for AQ pulping technologies have been improved over the past decade. First, the cost of AQ has decreased substantially over the past few years, as the AQ pulping patents have lapsed, new inexpensive off-shore AQ production facilities have been established, and several new AQ derivatives have been reported in the literature that exhibit improved activity [68].

## 5.3. Organosolv Pre-Treatment

With that focus, organosolv pulping processes have very few advantages compared to the traditional kraft process. However, organosolv processes may turn out to be potential as pre-treatment technologies for bioethanol production, aiming at delignification and opening of the cell wall matrix [48].

According to reference [69], a successful organosolv process for the replacement of the kraft process has the following characteristics (even if some of them are specific to paper grade pulp, most are relevant also in case of bioethanol as the target product): 1) totally sulfur free; 2) solubilization of most of the lignin with little loss of cellulose and hemicelluloses; 3) moderate temperature, pressure and pulping times; 4) efficient and simple chemical recovery system; 5) no environmental problems; 6) the optimal size of the process smaller than that of the kraft process; 7) applicable to broad selection of raw materials; 8) recovery of valuable by-products; 9) high pulp quality; 10) good bleachability without chlorine chemicals; 11) high pulp yield; 12) low energy consumption of the process; and 13) closed chemical cycle of the process.

The search for pulping processes that could fulfill the requirements listed above has led to the development of several organosolv methods capable of producing pulp with properties near those of kraft pulp. The Organocell (methanol), ASAM (methanol), Acetosolv (acetic acid) and Milox (peroxyformic acid) processes were tested either in pilot or full scale, but none of them led to continued production [70]-[75].

Nonwoody raw materials differ from wood in their chemical composition, especially concerning their high silica content which is problematic in kraft or soda cooking; during the delignification process, the silica present in the raw material dissolves into the cooking liquor, which has led to difficulties in the recovery chemicals system. Organosolv processes are therefore especially interesting for nonwoody raw materials, and at least the formic acid process has been commercialized [76]. The Lignofiber (LGF) process is a novel versatile organosolv method that fulfills most of the requirements listed previously, being applicable for annual plants as well as wood raw materials [77].

### 5.4. Enzymes as Potential Cell-Wall Deconstructing Biocatalysts

Several enzyme types are involved in natural biomass degradation [78] [79]. Many lignocellulose degrading enzymes work in a hydrolytic reactions system (mainly acting on hemicelluloses), while others works as oxidoreductive ones (mainly acting on lignin) [79]. Cell wall decay is a sequential process where lignin removal is generally the first and rate limiting step. When the cell wall protection due to the recalcitrant lignin is removed, the polysaccharides are susceptible towards attack by hydrolytic enzymes; however, an initial degradation of lignin-polysaccharide linkages can be produced in some cases [38] [80].

## 5.4.1. Lignin-Degrading Enzymes

Lignin is an aromatic heteropolymer consisting of three monolignols, methoxylated to various degrees: coniferyl alcohol, sinapyl alcohol and p-coumaryl alcohol. These monolignols are incorporated into lignin in the form of guaiacyl (G), syringyl (S) and p-hydroxyphenyl (H), respectively. Lignin are natural glues that bind tightly to polysaccharides, complicating our understanding of its native structure; the lignin-polysaccharide presents covalent and other linkage types, whose nature varies in different plants [81] [82] resulting in different susceptibilities towards enzymatic deconstruction. Whereas direct linkages between polysaccharide and lignin exist in wood, this association in most nonwoody angiosperms involves *p*-hydroxycinnamic acids forming lignin-polysaccharide bonds [83]. In first studies on enzymatic degradation of lignin during the 1970s the involvement of a variety of enzymes was considered [84]. This was because of the structural complexity of the lignin polymer (formed by different units and inter-unit linkages) that, according to the classic concept of enzyme-substrate specificity, suggested different enzyme types acting on the different substrate substructures. However, the main lignin degrading organisms (the so-called white-rot fungi because of the whitish color of the cellulose enriched substrate after lignin removal) have solved the problem of lignin heterogeneity just in the opposite way. The current evidence indicates that they have developed a unique degradation mechanism based on high redox-potential oxidoreductases, which often act in combination with redox mediators increasing their oxidative power [85]. These enzymes include laccases (fungal phenoloxidases) and ligninolytic peroxidases, the so-called lignin peroxidase (LiP), manganese peroxidase (MnP) and versatile peroxidase (VP), acting in synergy with H<sub>2</sub>O<sub>2</sub>-producing oxidases, among other enzymes [86]-[88].

Lignin degradation by the above oxidoreductases has been defined as an "enzymatic combustion" [89] be-

cause of the unspecific enzymatic attack, which is directed towards the benzenic ring of the lignin substructures. This results in a variety of subsequent side-reactions of the aromatic cation radical formed resulting in ether and side-chain breakdown (leading to depolymerization), demethoxylation, and even aromatic-ring opening. Because of the bulky nature of lignin, the lignin-degrading peroxidases have developed a noteworthy mechanism enabling oxidation of the polymer at the enzyme surface by an exposed protein radical, and subsequent long-range electron transfer to the peroxide-activated heme cofactor (that is located in a central pocket) [85] [90]. An alternative is the use of redox mediators, simple compounds that form diffusible stable radicals once oxidized by the enzyme, acting at distance and even when the access to the lignin polymer is limited by steric hindrances (such as the compact structure of the plant cell-wall in undecayed materials). These mediators have been largely investigated for their use in combination with laccases in different industrial applications including paper pulp delignification in totally chlorine free industrial-type bleaching sequences [91].

#### 5.4.2. Hemicellulose-Degrading Enzymes

Hemicelluloses vary in their constitutive monosaccharide units, as well as in their branching and acetylation degrees. Therefore, a variety of glycosidases and esterases are involved in hemicellulose degradation [92]. The former are often 1,4- $\beta$ -glycosidases, since 1,4-linkages predominate among hemicellulose units, whereas the latter mainly include acetyl-esterases releasing acetic acid from acetylated units, but also feruloylesterases breaking down the lignin-polysaccharide bridges mentioned above. Hemicellulose deacetylation can affect the substrate specificity of glycosidases involved in subsequent depolymerization [82], and is also of technological interest in biomass pre-treatment since the acetic acid released can contribute to lignin depolymerization by solvolysis reactions (as occurs in the organosolv processes).

Pectinases are important for cell-wall deconstruction in non-fully lignified tissues, where pectin in primary wall acts as the intercellular gluing agent. Since hemicelluloses occupy an intermediate position in the lignocellulosic matrix, being covalently-linked to lignin (directly or via cinnamic bound) and establishing hydrogen bond interactions with cellulose, its degradation often results in cell-wall deconstruction and separation of its two main constituents, cellulose and lignin. In a similar way, xylanases are used for years in the industrial bleaching of paper pulp [88], since xylan hydrolysis facilitates the removal of the residual lignin remaining in the paper pulp.

#### 5.4.3. Cellulose-Degrading Enzymes

Although cellulose is a simple polymer formed by cellobiose units, several enzymes (different 1,4- $\beta$ -glucosidases) are involved in its hydrolytic breakdown [93]. These include the so-called endoglucanases breaking down internal glycosidic bonds in the cellulose polymer, a process that results in a decrease of molecular mass (polymerization degree). On the contrary, 1,4-cellobiohydrolases (exocellulases) release the cellobiose units from cellulose chain-ends, which are hydrolyzed by cellobiases ( $\beta$ -glucosidases) yielding free glucose. The action of the latter enzymes is facilitated by the depolymerizing action of the above endoglucanases that increase the number of chain-ends susceptible of attack by exocellulases. The action of cellulases is facilitated by the existence of conserved cellulose-binding domains in their molecular structure. The bacterial cellulosomes are suggestive examples of complex multienzymatic systems including different carbohydrate hydrolases and carbohydratebinding modules on a common scaffold for their concerted action on a lignocellulosic substrate [94]. Crystallinity is a main bottleneck for cellulose hydrolysis by cellulase complexes and some proteins lowering the crystallinity degree have been also described in cellulolytic organisms, such as fungal swollenins [95]. Moreover, an extended attack on crystalline cellulose, without the previous removal of lignin, is produced by brown-rot fungi.

In the same way that the unique ligninolytic oxidoreductases produced by white-rot fungi are of enzymes of interest as industrial biocatalysts, the so-called brown-rot fungi have developed a noteworthy mechanism for the selective removal of wood carbohydrates leaving a lignin-enriched residue [96] that could be applied in ligno-cellulose biorefineries. Recent studies, including the complete sequencing of the first genome of a brown-rot fungus at the DOE Joint Genome Institute (www.jgi.doe.gov) have shown that their ability to use lignin, without a previous removal of the lignin barrier, is based on the enzymatic generation of hydroxyl radical in the so-called Fenton reaction (consisting of peroxide reduction by ferrous iron). The hydroxyl free radical is one of the strongest oxidizers involved in biological reactions, depolymerizing cellulose even in its crystalline state. These findings suggest the feasibility of a biomimetic approach based on Fenton chemistry to improve polysaccharide hydrolysis for the subsequent production of bioethanol and other products from the lignocellulosic biomass.

## 5.4.4. Other Enzymes

In addition to the above enzymes, other enzymes with a potential interest in cell-wall deconstruction participate in the natural decay of lignocellulosic biomass. These include, among others, lipases hydrolyzing triglycerides representing a significant percentage of lipophilic extractives, and esterase and glycosidase hydrolyzing sterol esters and glycosides also present in this fraction [97]-[100]. Proteins also represent a minor fraction in lignocellulosic materials (together with the water-soluble material containing free sugars), and proteases (peptidases) hydrolyzing the amide type peptidic bonds can be of interest in the isolation of cell wall constituents [100].

# **6.** Conclusion

Biofuels e.g. bioethanol, are important in the future because they replace petroleum-based fuels, and they can be obtained from renewable sources such as sugar cane, corn, straw, grasses and wood. There are currently several different levels of system integration for biofuel production which add to their sustainability, both economically and environmentally. The pulp industry appears as a very interesting opportunity, since it already has efficient process to separate the lignin and extractives fractions from the polysaccharides; this cracking is very suitable and necessary for biorefinery applications. Efficient enzymatic conversion of lignocellulosic polysac-charides into sugars is a key technology in the future biorefineries and is currently the subject of intensive research.

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