The Cellulolytic Bacteria *R. albus* for Improving the Efficiency of Microbial Fuel Cell

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Abstract

The current study has been undertaken to examine the beneficial effect in the power output of a microbial fuel cell (MFC) by adding cellulolytic bacteria *Ruminococcus albus* (*R. albus*) into the anodic chamber. Mediator-less H-type MFCs were set up where the anode chamber contained anaerobic digestor microorganisms as inocula on finely ground pine tree (Avicel) at 2% (w/v) and the cathode chamber of 10mM phosphate buffered saline conductive solution, both separated by a cation exchange membrane. The functioning of the MFCs for generation of electrical power and the amounts of gaseous byproducts was monitored over a 9-day period. The addition of cellulolytic bacteria caused an increase of average power density from 7.9 m W/m² to 19.5 m W/m², about 245% increase over a 9-day period. For both groups of MFCs; with *R. albus* and the control, the head space gases collected were methane and CO₂. While the methane: CO₂ ratios were found unchanged at 1.7:1 throughout the 9 days of operation, the total gas production increased from 248 mL to 319 mL due to the presence of *R. albus* addition. This study confirms that whereas the biocatalytic activity of anode microbial population determines the energy production, the addition of external cellulolytic bacteria into anode microbial population can improve and extend the biomass utilization.

Keywords

Microbial Fuel Cell (MFC), Cellulolytic Bacteria, Microorganism, *R. albus*
1. Introduction

All over the world, fossil fuels such as petroleum, coal, and natural gas have served as the main energy resources for industrialization and economic growth for the past century [1], and represent 79.4% of the global primary energy use in 2001 [2]. While 418 exajoules \((10^{18} \text{ J})\) of energy was used and total sales of energy carriers amounted to an astounding US $2 trillion world-wide in 2001, more than 2 billion people still cannot access affordable energy services [2]. Fossil fuel reserves may run out in just over 100 years [3]. Not to disregard, the use of fossil fuels negatively impacts the environment due to the emission of greenhouse gases including CO₂, methane, and CO, which cause global warming and pollution [4]. For these reasons, greater efforts are currently being undertaken worldwide to develop technologies that generate clean, sustainable energy sources that would replace and/or displace fossil fuels [5].

As one of the most abundant renewable resources, cellulosic biomass is particularly attractive because of its relatively low cost, plentiful supply [6] [7], and neutral carbon balance [8]. The U.S. Departments of Agriculture and Energy estimated the annual availability at 1.3 billion dry tons of biomass feedstock in the United States, which could replace 30% or more of the country’s present petroleum consumption. It should be noted that cellulose is a significant component in the annual production of 250 million tons municipal solid wastes and 40 billion cubic meters wastewater [9]. Depending on the end-use application, cellulosic biomass could be converted to a variety of energy carriers such as ethanol [10], biodiesel [11], and hydrogen [12], as well as to electricity indirectly derived from cellulose by coupling cellulolytic, fermentative hydrogen production with the catalytic oxidation of hydrogen at a fuel cell anode [8].

Representing an alternative method of renewable energy recovery, the direct conversion of biomass including organic waste to electricity using a microbial fuel cell (MFC) or to hydrogen, using a microbial electrolysis cell (MEC) offers the potential of clean and distributed energy production [13]. Using electrochemically active microorganisms as biocatalysts, MFCs are bioelectrochemical reactors that convert organic materials directly into the electricity [13], and they are endowed with tremendous electron donor versatility. These include simple substrates such as glucose, acetate, and lactate [14] [15] [16] [17]; complex substrates such as municipal and industrial wastewaters [18] [19]; even steam-exploded corn stover hydrolysate [20] and finally, cellulose [1] [21] [22] [23] [24].

Many types of bacteria have been found to be electrochemically active however, none of them is known to show a cellulolytic activity to generate electrons directly, and most of them require hydrolysis products of cellulose as electron donors [14] [25] [26] [27]. Exogenous redox mediators for electron transfer to the electrode are not desirable because these electron shuttles are toxic and need to be replenished periodically. Due to the lack of an isolated microbe that can both hydrolyze cellulose and reduce solid extracellular electron acceptors, one strategy to directly produce electricity from cellulosic biomass would involve a
synergistic consortium of polymer-degrading fermentative microorganisms and fermentation products by utilizing electrochemically active microorganisms. Microorganisms that have been tested as biocatalysts for use in MFCs include pure or defined co-culture of obligate and facultative anaerobic bacteria [1] [14] [22] [25], and mixed cultures from sea floor sediments [28] [29], municipal and industrial wastewater or anaerobic digester [30] [31] [32], soil [1] [33] and rumen microbiota [24].

The rumen microbiota contains both strict and facultative anaerobes, which effectively hydrolyze cellulose and conserve energy via anaerobic respiration or fermentation, have been used for enhancing anodic efficiency [24] [34]. Microorganisms in anaerobic digesters have shown electrochemical activities that successfully transferred electrons to anode in MFCs [35] [36] [37] [38] [39]. Including rumen contents or anaerobic digester contents, studies have been carried out with mixed microorganism as anolyte inoculum to investigate single resource of cultures as both cellulolytic and electron transfer microorganisms. Studies of biomass degradation and electron transfer to electrode are conducted by different microorganisms, where more than one resource of mixed microorganisms may be selected for each symbiotic and synergetic role in MFCs.

We hypothesized that the combination of strong cellulolytic microorganisms and well established electrochemically active microorganisms in anode chamber would ferment cellulosic biomass more efficiently, and subsequently generate more electricity in MFCs. In the current study, we established MFCs consisting of anaerobic digester microorganisms as anolyte and cellulose as electron donor, and further by supplementing with culture of cellulolytic rumen bacteria, R albus [40], and investigated whether this would result in an increase of cellulose fermentation and whether the higher cellulose degradation would directly translate to an enhanced electricity generation in MFC.

2. Materials & Methods

2.1. Microorganisms and Culture Media

The anaerobic digester fluid used as MFC anode chamber inoculums was collected from a dairy farm with the following treatment: Under flushing of CO2 gas through heated copper column, the fluid was filtered through 4 layers of cheesecloth and glass wool, and then bubbled with CO2 gas until being transferred to MFCs.

A stock culture of R. albus stain 7 (ATCC® 27210™) was transferred to an anaerobically prepared medium containing cellobiose, 1 g; KH2PO4, 0.48 g; K2HPO4, 0.48 g; (NH4)2SO4, 0.48 g; NaCl, 0.96 g; Trypticase, 5.0 g; yeast extract, 1.0 g; isobutyric acid, isovaleric acid, and DL-2-methylbutyric acid, 0.1 ml of each; cysteine hydrochloride, 0.5 g; CaCl2·2H2O, 0.13 g; MgSO4·7H2O, 0.2 g; Na2CO3, 4.0 g; sodium fumarate, 1.0 g, and resazurin, 1.0 mg per 1 L of medium volume with distilled deionized (dd) H2O. After 48 h incubation at 39°C, an aliquot of 0.5 mL culture was transferred to 10 mL of the same fresh medium and
incubated for 48 h at 39°C. Thereafter, 1 mL of the subculture was inoculated to anode chamber of the MFCs.

Phosphate buffered saline (PBS) was prepared by dissolving NaCl, 8 g; KCl, 0.2 g; Na₂HPO₄, 1.44 g; and KH₂PO₄, 0.24 g in 800 ml ddH₂O and adjusting the pH to 7.5 and volume to 1 L with ddH₂O. PBS was autoclaved at 121°C for 30 min and stored at room temperature until being transferred to MFCs.

2.2. Microbial Fuel Cells

Mediator-less two chamber H-type microbial fuel cell was constructed using two 125 mL-volume glass jars, as anode and cathode chambers, adjoined at branched tubular bridge, and a proton exchange membrane (CMI-7000S, Membranes International Inc., NJ) clamped between tubular bridges, separating them.

A 2 g cellulose (Avicel PH-101, 11363 Sigma-Aldrich, MO) and 100 mL of anaerobic digester fluid collected from a dairy farm were transferred to the anode chamber and suspended by agitation. Graphite stick (12 cm²) connected with copper wire was placed in the middle of anode chamber and the chamber was closed with butyl rubber stopper. A 100 mL PBS was transferred to the cathode chamber and a graphite stick (12 cm²) connected with copper wire was placed in the middle. Rubber stopper was used to cap the cathode chamber but was made open to air through a tubing on the stopper. Anode and cathode were connected externally with a copper wire and a load resistor (300 ohm). MFCs were operated in a water bath at 39°C for 9 d prior to the treatment inoculation.

After 9 d of MFC operation, before treatment inoculation, current density for MFCs was measured to be 176 ± 6.5 mA/m². One mL of R. albus, grown in a cellobiose medium for 48 h, was inoculated into anode chambers of the treatment group MFCs, and 1 mL of pure cellobiose medium was added to anode chambers of the control group MFCs. Anode chamber stoppers were open to the atmosphere to equalize pressure and remove the headspace gas. 2 L-volume balloons were connected to each anode chamber to collect gases produced.

2.3. Measurements and Calculation

MFC voltage across an external resistor, end point potential, and current were measured daily using a multimeter from d₀ to d₉. The power density was calculated according to \( P = I \times V / A \), with \( I = \) current, \( V = \) voltage, \( R = \) external resistance, and \( A (m²) = \) the projected area of the cathode.

On d₉, balloons connected to anode chambers were collected and total volumes of fermentation gas produced were measured. CO₂ and methane profile was analyzed using an Agilent gas chromatograph equipped with a thermal conductivity detector and a stainless steel packed (12390-U, Supelco, Sigma-Aldrich, MO).

2.4. Statistical Analyses

Effects of R. albus addition to anaerobic digester fluid in anode chamber of MFC
on electricity generation, fermentation gas production and gas composition were analyzed using the one-way ANOVA procedure of JPM 12.2.0 (SAS Institute Inc., NC) and when the effect was significant ($P < 0.05$), treatment means were separated using students’ t-test. Significance was declared at $P < 0.05$.

3. Results and Discussion

3.1. Voltages

MFCs consisting of 2% w/v Avicel and 100 mL of anaerobic digester fluid were constructed and stabilized for 9d and power density was measured to be 11.2 ± 0.86 mW/m² prior to applying the treatments. It was noted in control group that the voltage across resistor and endpoint potential (open circuit voltage) showed a decrease ($P < 0.05$) on d2 and fluctuated thereafter with time courses (Table 1). The control group as received had 1% v/v of pure medium containing 0.1% cellobiose without $R$. albus, such that it was assumed that a slightly higher voltage on d0 and d1 (compared to the rest of operation period) might be induced by the spike of 0.001% cellobiose in the final volume of anolyte.

The Voltage across resistor and endpoint potential in the presence of $R$. albus group increased gradually after 3d and then reached the highest at d8 (Table 1). $R$. albus MFCs had induced a greater ($P < 0.05$) voltage across resistor and endpoint potential than control MFCs after d2 except on d4 for open circuit voltage (Table 1).

3.2. Gases Composition

Supplementation of $R$. albus to anaerobic digester fluid increased overall gas

| Table 1. Closed circuit voltage across 300 ohms resistor and terminal. |
|-----------------------------|-----------------------------|
| Voltage across resistor (300 ohms), mV | Open circuit voltage, mV |
| Day | Control | $R$. albus | SEM | $P^2$ | Control | $R$. albus | SEM | $P^2$ |
| 0 | 64.0| 60.0 | 2.24 | 0.3333 | 206.0 | 185.5 | 10.25 | 0.2930 |
| 1 | 61.0 | 65.0 | 2.55 | 0.3828 | 183.0 | 190.0 | 7.78 | 0.5897 |
| 2 | 50.0 | 69.5 | 2.85 | 0.0402 | 152.0 | 219.0 | 6.52 | 0.0184 |
| 3 | 57.5 | 80.5 | 2.69 | 0.0263 | 182.5 | 236.0 | 7.56 | 0.0377 |
| 4 | 53.0 | 86.5 | 3.48 | 0.0209 | 163.5 | 262.5 | 17.87 | 0.0594 |
| 5 | 49.0 | 93.0 | 1.58 | 0.0026 | 144.0 | 268.5 | 2.85 | 0.0010 |
| 6 | 50.5 | 93.0 | 4.26 | 0.0195 | 153.5 | 287.5 | 11.93 | 0.0155 |
| 7 | 49.0 | 89.5 | 2.26 | 0.0062 | 155.5 | 265.5 | 4.27 | 0.0030 |
| 8 | 44.5 | 100.5 | 2.69 | 0.0046 | 133.0 | 295.5 | 5.06 | 0.0019 |
| 9 | 49.5 | 90.0 | 3.02 | 0.0109 | 150.0 | 272.0 | 8.28 | 0.0091 |
| SEM | 2.60 | 3.29 | 7.31 | 10.77 |
| $P^2$ | 0.0028 | <0.0001 | 0.0010 | 0.0002 |

$^{a,b,c,d}$Means with different superscripts differ, $P < 0.05$. $^1$Standard error of means. $^2$P-value; probabilities that treatments or day effects are not significant.
production, but did not change the gas composition (Figure 1). The total gas produced in anode chambers for 9 d after treatments amounted to 249 and 319 mL for control and R. albus MFCs, respectively (Figure 1). Gas production implies that the fermentation of electron donors occurred during the electricity generation, suggesting that fermentation was greater \((P = 0.0351)\) in R. albus MFCs. While methane \((P = 0.0548)\) and CO\(_2\) \((P = 0.0478)\) volumes were greater in R. ablus MFCs, the methane to CO\(_2\) ratios remained same for both MFCs \((P = 0.9381)\) with a ratio of 1.7:1.

The basic process for glucose in anode chamber is decomposition into CO\(_2\), proton and electron. The electrons and the protons move to cathode via the external electrical circuit and the exchange membrane, respectively, then reduce oxygen and produce water. Overall scheme in whole MFC is \(\text{C}_6\text{H}_{12}\text{O}_6 + 6 \text{O}_2 \rightarrow 6 \text{CO}_2 + 6\text{H}_2\text{O} + \text{Electrical energy} \) [41]. Both acetoclastic methanogenesis and hydrogeno-trophic methanogenesis require exogenous energy consumption [42] and reduce the flow of proton and electron to cathode, therefore methane production is regarded an inefficient process which detracts from electricity generation. In the current study, a similar pathway of cellulose fermentation can be deduced from the production rate of methane to CO\(_2\), which was not different between the R. albus and the control MFCs, and this implies that production of methane is a characteristic of anaerobic digester process.

3.3. Power Density

Power density (power normalized to the surface area of the electrodes) is a critical parameter to determine the bioelectrochemical performances of MFCs [43]. Power density in the control MFCs decreased \((P < 0.05)\) on d2 and thereafter remained steady at between 5.5 and 7.8 mW/m\(^2\) (Figure 2) while in R. albus MFCs, power density increased \((P < 0.05)\) gradually after d3 registered in the range of 20.9 to 28.1 mW/m\(^2\). When a sole rumen fluid was used as anolyte [24],

![Figure 1](https://example.com/figure1.png)

**Figure 1.** Components and total volume of gases produced in the anode chamber of microbial fuel cells for 10d incubation. Methane, CO\(_2\), and total gas are presented as means of duplicate \((n = 2)\) with standard deviation.
Figure 2. Power densities over 9 day incubation. Power densities are presented as means of duplicate (n = 2) with standard error of mean for control and R. albus treatment.

55 and 26.7 mW/m² were reported as the highest and stable power density, respectively. However, in their setup, aerobic potassium ferricyanide solution (50 mM K₃Fe(CN)₆) had been used as the catholyte to enhance oxygen reduction, so that the performance of our MFCs will not stand for a direct comparison. Besides, the performance can be restricted by the large internal resistance of H-type fuel cells with long distance between the anode and cathode and small surface area of the cation exchange membrane and hence, the power outputs need to be properly normalized [13].

Continuous power generation in control MFCs reflects the ongoing fermentation in anode chamber and the flow of electron and proton to cathode chamber. Gas production and power generation throughout the experimental period in the control MFC implies that anaerobic digester fluid might include not only electrochemically active but also cellulose degrading microorganisms. In accordance with the greater gas production, R. albus fermented and lysed cellulose more efficiently than the control, and provided more electron donors, which were available to electron transfer microorganisms on anode and resulted in greater power generation.

4. Conclusions

Because of its abundance and carbon neutral characteristics, cellulosic biomass is a desirable ingredient for biofuel production. In the current study, anaerobic digester fluid was employed as an anolyte in MFCs, which generated electricity from finely ground pinetree (cellulose). Addition of R. albus at 1% v/v into the anolyte increased gas production and power generation. These results suggest that addition of R. albus or other exogenous cellulolytic microorganisms to anaerobic digester MFC may degrade cellulosic biomass more rapidly such that generate electricity more efficiently. While this study draws a close similarity
with that of Rismani-Yadzi et al. (2007), the difference in the composition of the catholyte is a distinct difference and demonstrates how the efficiency of the MFCs can be optimized vis-a-vis avoidance of toxic components such as potassium ferricyanide.

For a large scale development, although R. albus served as an important prototype, many other similar exogenous cellulolytic microorganisms should be screened in order to establish common structural characteristics and mechanistic pathway that facilitate efficient electrode processes.

References


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