Crystallization and Characterization of a New Nonlinear Optical Crystal: L-Proline Succinate (LPS)

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ABSTRACT

In this analysis, the single crystal of L-Proline Succinate (LPS) has been successfully synthesized and the purity of material has been increased by repeated recrystallization process. Single crystal was grown by adopting the method of growing in a slow evaporation solution using water as solvent at room temperature. The LPS single crystal has been synthesized by taking equimolar quantity of L-Proline and succinic acid, by mixing them thoroughly using deionized water. The prepared concentrated solution was placed in an undisturbed condition, and the solution was inspected regularly. The single crystal has been harvested over a period of 1 month. The same crystal was characterized by different techniques for finding its suitability for device fabrications. The grown crystal was characterized by Single crystal XRD, Powder XRD, FTIR, UV-vis-NIR, DTA/TGA and SHG analyses, respectively. The observed results from various characterization show the suitability for NLO application. The second harmonic generation of this grown crystal was checked using Kurtz Perry technique which showed positive results. The UV cut-off wavelength and the decomposition temperature of this grown crystal were found to be good when compared with the existing organic crystals.

Keywords: Slow Evaporation; Crystal Growth; X-Ray Diffraction; Fourier Transform Infrared Spectroscopy; Second Harmonic Generation; LPS Crystal

1. Introduction

One of the most important applications of NLO materials is their use for fast data transfer, combined with a very high Signal-to-Noise ratio, even over long distances. In recent years, different applications of NLO and photorefractive materials have been developed, for example, optical frequency conversion, electro-optical modulation, dynamic holography, optical writing and optical guiding of laser beams [1]. It is seen that L-proline and (4R)-hydroxy-L-proline derivatives, containing donor groups are chiral carriers [2]. The introduction of chirality by means of an asymmetrically substituted carbon should in addition respect the molecular features leading to a high nonlinear behaviour [3]. Proline and its derivatives are often used as asymmetric catalysts in organic reactions.

Only noncentrosymmetric alignment of the chromophores in the crystal lattice leads to an observable bulk second-order NLO response [4]. In order to obtain the adjustment of the nonlinear efficiency/transparency, based on the molecular engineering and crystal engineering approach, it is tried to develop a new method to design organic nonlinear optical second-harmonic generation materials such as organic inclusion complex [5]. In this study, the crystal growth of a new NLO crystal of L-Proline succinate by slow evaporation technique and its characterization along with its optical properties is reported.

2. Experimental Details

Equal proportions of L-Proline and succinic acid were taken and were dissolved separately in deionized water. Then the solution of L-Proline was poured into the dissolved succinic acid mixture. The solution thus arrived was filtered twice to remove dust particles and undissolved materials. The reaction takes place between L-Proline and succinic acid (acid-based) through hydrogen transfer. Thus formed ionic compound of L-Proline succinate is represented in the following equation:

![Chemical structure of L-Proline Succinate](image)

The saturated solution was maintained in the undisturbed condition and the beaker was covered by polythene paper. Few holes were made on the polythene cover for slow evaporation. By adopting the solution growth method,
single crystal of L-Proline succinate (LPS) was grown from supersaturated solution at room temperature. Then this solution was periodically inspected and from the 20th day onwards the crystal started growing and it was permitted to grow for another 10 days in order to get a nominal size suitable for characterization. The single crystal of LPS with dimensions of 8 mm × 5 mm × 10 mm was thus obtained. The L-Proline interacts with succinic acid through a single N-H-O hydrogen bond. A single crystal of LPS which has been grown by this process is shown in Figure 1.

**Characterization**

The lattice parameters and the crystal systems have been determined using single crystal X-ray diffraction analysis (Model: Bruker AXS Kappa APEX II single crystal CCD diffractometer). The functional groups presented in the LPS compound have been identified by Bruker IFS 66V model FTIR Spectrometer using KBr pellet technique in the region 400 - 4000 cm⁻¹. Optical behaviour of LPS was measured by Perkin Elmer Lambda 35 UV-VIS-NIR Spectrophotometer in the wavelength range of 190 - 1100 nm. The thermal stability of LPS was studied by thermo gravimetric analysis (TGA) and differential thermal analysis (DTA) by using SDT Q600 V8.3 Build 101 thermal analyzer instrument ranging from room temperature to 1100°C at a heating rate of 20°C per minute under nitrogen atmosphere.

### 3. Results and Discussion

#### 3.1. Single Crystal X-Ray Diffraction Analysis

Single crystal X-ray diffraction studies were carried out on the grown crystals. The X-ray data were collected using X-ray diffractometer (Model: Bruker AXS Kappa APEX II single crystal CCD). The observed results indicate that the crystal belongs to monoclinic crystal system and the determined unit cell parameters are a = 5.07 Å, b = 8.84 Å, c = 5.48 Å, α = 90°, β = 91.60°, γ = 90° and V = 246 Å³.

#### 3.2. FTIR Spectroscopy

The functional groups presented in the LPS compound have been identified by Bruker IFS 66V model FTIR Spectrometer using KBr pellet technique in the region 400 - 4000 cm⁻¹. The FTIR spectrum of title compound is shown in Figure 2. The peaks obtained are 3419 cm⁻¹, due to stretching vibration of CH and the peak at 1600 cm⁻¹ is due to the stretching vibration of C=O. The bands appeared at 793 cm⁻¹ is assigned unambiguously to the wagging of NH2 modes. The OH stretching vibrations is assigned in the range of 2565 cm⁻¹. The peak at 1398 cm⁻¹ is due to the symmetric stretching of COO⁻. These assignments are also supported in the literature [6-10]. The observed bands along with their vibrational assignments are given in Table 1.

#### 3.3. UV-Visible Spectroscopy

Good optical transmittance and lower cut-off wavelengths are very important properties for NLO crystals. Optical behaviour of LPS was measured by Perkin Elmer Lambda 35 UV-VIS-NIR spectrophotometer in the wavelength range of 190 - 1100 nm. The recorded spectrum is shown in Figure 3. The crystals are broadly transparent possessing a transmission of greater than 90% for light with incident wavelengths from 236 - 1100 nm. The UV transparency cut-off wavelength of LPS crystal occurs at 204 nm which is better than L-Prolinium tartrate and 4-phenylpridinium hydrogen squarate [11,12]. It is observed
that in the LPS crystal, there is high transmittance in the far ultraviolet, visible and infrared region. Hence, the title compound may be used for the nonlinear optical applications in the above mentioned wavelength range.

3.4. Second Harmonic Generation
The SHG of the crystal was checked using the powder SHG technique developed by Kurtz and Perry [13]. A Q-switched Nd:YAG laser beam of wavelength 1064 nm, with beam energy of 4.5 mJ/pulse, and pulse width of 8 ns with a repetition rate of 10 Hz were used. The grown single crystal was crushed to fine powder and then packed in a micro capillary of uniform bore and exposed to laser radiations. The 532 nm radiation was collected by a monochromater after separating the 1064 nm pump beam with an infra-red blocking filter. The second harmonic radiation generated by the randomly oriented micro crystals was focused by a lens and detected by a photo multiplier tube (Hamamatsu R2059). The second harmonic generation is confirmed by the emission of green light and its efficiency is found to be 23% of that of KDP crystal.

3.5. Thermal Analysis
The thermal behaviour of LPS had been studied by thermo gravimetric analysis (TGA) and differential thermal analysis (DTA) using SDT Q600 V8.3 Build 101 thermal analyzer instrument, ranging from room temperature to 1100 °C at a heating rate of 20°C per minute under nitrogen atmosphere. TGA-DTA curve of L-Proline succinate is shown in Figure 4. From the TGA curve, the material starts to decompose around 160°C, which is also confirmed by the peak which appears at 154°C in the DTA curve. The thermal stability of LPS is low compared to L-Proline (205°C) and succinic acid (240°C). The TGA curve shows a major loss of weight and two losses of smaller weight. The major weight loss of 86.58% is observed between 169.91°C and 243.16°C. The decomposition of L-Proline succinate (87.53%) leads to the major loss of weight in the above mentioned temperature region. Further, a small loss of weight of 10.24% is observed, which may be due to the decomposition of C₂H₂ (11.16%). During the next stage of decomposition, fraction amount of hydrogen may be decomposed. A residue of 0.9018% which may be due to presence of some fraction of carbon molecule is observed. The thermal stability of the LPS single crystal is more than glycine nitrate [14] and is lower than gamma glycine and glycine acetamide [13,15].

4. Conclusion
We have synthesized a new non-linear optical crystal with an interesting hydrogen bonding network that holds together the L-Proline and succinic acid molecules. The grown crystals are characterized by different instrumental techniques. The single crystal XRD studies prove that the grown LPS crystals belong to monoclinic crystal system. The particle size of the grown crystal is characterized by Powder XRD analysis. The presence of the functional groups of the grown crystal has been confirmed by FTIR analysis. From the optical transmittance spectrum, it is observed that there is high transmittance in the far ultraviolet, visible and near infra red regions. The UV transparency cut-off wavelength of LPS crystal occurs at 204 nm. The Kurtz Perry technique for second harmonic generation has showed positive results. It is well known that the DTA and TGA studies reveal that the crystal is thermally stable up to 160°C.

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