Surface Modification on Ti-30Ta Alloy for Biomedical Application

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ABSTRACT

Titanium and titanium alloys are currently being used for clinical biomedical applications due to their high strength, corrosion resistance and elastic modulus. The Ti-30Ta alloy has gotten extensive application as the important biomedical materials. The substrate surface of the Ti-30Ta alloy was altered either by chemical or topographical surface modification. The biocompatibility of an implant is closely related to its surface properties. Thus surface modification is one of effective methods for improving the biocompatibility of implants. The development status of biomedical materials has been summarized firstly, as the biomedical application. In this study Ti-30Ta alloy surface was investigated as-cast- ing (Group 1) modified with alkaline and heat-treatments in NaOH with 1.5M at 60°C for 24 hrs (Group 2), alkaline and heat-treatments with SBF-coatings by immersion in NaOH and SBFX5 for 24hrs (Group 3), anodization process was performed in an electrolyte solution containing HF (48%) and H2SO4 (98%) with the addition of 5% dimethyl sulfoxide (DMSO) 35V for 40 min (Group 4) and ion beam etching with 1200 eV ions with a beam current of 200 mA for a 3 hrs etch (Group 5). SEM was used to investigate the topography, EDS the chemical composition, and surface energy was evaluated with water contact angle measurement. SEM results show different structure on the surface for each group. EDS spectra identified similarity on Group 1, 4 and 5. The results indicate for group 2 an amorphous sodium tantalate hydrogel layer on the substrate surface and for group 3 the apatite nucleation on substrate surface. The Group 4 shows unorganized and vertically nanotubes and Group 5 shows a little alteration in the topography on the substrate surfaces. Overall the contact angle shows Group 5 the most hydrophobic and Group 4 the most hydrophilic. The study indicates Group 3 and 4 with potential for biomedical application. The next step the authors need to spend more time to study group 3 and 4 in the biomedical sciences.

Keywords: Biocompatibility; Ti-30Ta Alloy; Alkali Treatment; Heat Treatment; Simulated Body Fluid; Anodization Process; Ion Beam Etching

1. Introduction

Metallic materials have been used as implantable for orthopedic and dental implants. Materials such as steel, cobalt alloys and titanium and titanium alloys are currently being used in clinical biomedical applications due to their high strength, corrosion resistance and elastic modulus [1-6]. However, these materials have recently been shown to exhibit ion release and poor physiological integration that may result in fibrous encapsulation and further biomaterial rejection [3,7,8]. In order to be a successful replacement for bone a biomaterial must be bio-compatible, corrosion resistant, exhibit a high strength, and a low elastic modulus [1-6]. Current approaches for enhancing the mechanical and biological properties of Ti alloys include add elements such as Nb, Ta, Zr, Mo, Hf and Sn [4,9-23]. Among these candidates, titanium (Ti) alloyed with tantalum (Ta) will be considered in this study due to it provides greatly improved mechanical properties which include fracture toughness and workability, shows improvement from both pure Ti and pure Ta [18,24-27]. In addition, using 30% of tantalum has been shown to produce a strong effect on both young’s modulus and the tensile properties of binary Ti-Ta alloys. Unique to other Ti-Ta alloys, the Ti-30Ta alloy provides...
a martensite “α” phase that produces a good combination of low modulus and high strength [9-13,21,24,25,28]. The interaction between the implant surface and the tissue plays an important role on the success of this implantable device [3,7,29]. Several studies have shown that by modifying the surface at a nanoscale or a microscale it can alter cellular response [20,30-32]. Studies have shown techniques such ion beam etching [33,34], heat and alkaline treatment [15,17,35], SBF coatings [16, 36-38] and anodization [39-41] to promote altered cellular response on Ti and Ti-alloys. One effective parameter to evaluate the biological response of the metallic biomaterials is investigating the wettability of the surface of this material due to the topography having effect on protein adsorption, platelet adhesion, blood coagulation and bacterial adhesion [42-45]. In this study, the substrate surface of the Ti-30Ta alloy was altered either by chemical or topographical surface modification. The Ti-30Ta alloy substrates were modified with alkaline and heat-treatments, alkaline and heat-treatments with simulated body fluid (SBF) coatings, anodization process and ion beam etching. Following techniques were used for characterizing all groups: scanning electron microscopy (SEM), energy dispersive spectroscopy (EDS) and contact angle analysis.

2. Materials and Method

2.1. Fabrication of Ti-30Ta Alloy

Ti (99%) and Ta (99%) were combined by a melting process in a high purity argon atmosphere; re-melted ten times and held in the molten state for 3 - 4 minutes due to the titanium and tantalum there are differing densities (Ti: 1953 K, 4.51 g/cm³ and Ta: 3273 K, 16.6 g/cm³) and the wide-phase differential (liquid and solid) in the binary Ti-Ta phase diagram [9-13,22,25]. All ingots were homogenized in a vacuum at 1100°C for 86.4 ks in order to eliminate the as-cast microscopic chemical segregation and the residual stress caused by plastic deformation on the elastic modulus. The alloy was then cold-worked by a rotary swaging process (rotation speed 1000 rpm) using a CNC lathe ZIL (Centur 30S, ROMY, Brazil) [45] and cut in discs. The Ti-30Ta alloy substrates were then separated into five groups for this study: Group 1: Ti-30Ta alloy substrates control, Group 2: Ti-30Ta alloy substrates with alkaline and heat-treated, Group 3: Ti-30Ta alloy substrates with alkaline and heat-treatments with SBF-coatings, Group 4: Ti-30Ta alloy substrates submitted to anodization process and Group 5: Ti-30Ta alloy substrates after etching (See Table 1).

2.2. Alkaline and Heat Treatments on the Ti-30Ta Alloy Surface (Group 2)

The control substrate surface was used as-fabricated (Group 1). Group 2 substrates were alkaline-treated by soaked in 5.0 ml of NaOH (aqueous solution) with 1.5M at 60°C for 24 hrs and dried at 40°C for 24 hrs [15-17, 19, 46]. The substrates were further heated to 300°C in an electric furnace (air atmosphere) at a rate of 5°C/min for 1 hr [16,17] and cooled to room temperature within the furnace.

2.3. SBF Coatings on Ti-30Ta Alloy Surface (Group 3)

Group 3 substrates were alkaline and heat treated [16,17] and incubation on 30 ml of SBFx5 with ion concentrations and pH is comparable to that of human blood plasma at 36.5°C. Previous studies have identified SBFx5 [38] solution to accelerated the deposition of biomimetic Ca-P. The SBFx5 was prepared using the protocol described in publish literature [38].

2.4. Anodization of Ti-30Ta Alloy Surface (Group 4)

Group 4 substrates were fabricated by anodization process using platinum as the cathode and the Ti-30Ta as the anode. The electrolyte solution composed of concentrate HF (48%) and H2SO4 (98%) in the volumetric ratios 1:9 with the addition of 5% dimethyl sulfoxide (DMSO), stirred constantly. The experiment was conducted in constant voltage of 35 V during 40 minutes at room temperature. The substrates were cleaned with isopropyl and dried with compressed air. The substrates were then sintered in an electric furnace with oxygen ambient at 530°C rate of 1°C/min for 3 hrs.

2.5. Ion beam Etching of Ti-30Ta Alloy Surface (Group 5)

Group 5 substrates were fabricated by etching the alloy surface to oblique angle oxygen ion beam by increasing the oxygenation of the near surface regions. Etching was done with a 16 cm ion source in a low-pressure environment (approximately 1.6 × 10^-4 Torr). Gas flow rates through the source and neutralizer were 20 sccm O₂ and 8 sccm Ar respectively. An energetic ion beam of 1200 eV ions with a beam current of 200 mA was used for a
hour etch. This beam consisted primarily of oxygen ions though it is possible that minute amounts of background Ar gas could diffuse into the source, ionize and be included. The substrates were placed on an inclined holder so the resulting angle of ion incidence was approximately 75 degrees from the surface normal [33].

2.6. Physical Characterization of Different Groups

The Ti-30Ta alloy substrates surfaces were examined before and after the surface modification. The surface topography of the substrates was characterized by coating the surface with 15 nm of gold and using a JEOL JSM-6500FESEM at a working distance of 10 cm and a voltage of 15 kV. The surface elemental composition of the Ti-30Ta alloy substrates was further characterized with energy-dispersive X-ray spectroscopy (EDS, JSM-6500F SEM). Wettability of the modified surfaces were determined by measuring water contact angle (FTA1000B Class, First Ten Angstroms, Inc.). A 2 µl droplet of distilled water was dropped on the surface. Immediately after this the droplet images captured using a camera. The image was then processed with the accompanying Fta32 software determine give contact angle and the droplet volume. All the studies were conducted for minimum 6 samples to ensure appropriate statistical variability.

3. Results and Discussion

The binary Ti-30Ta alloy was fabricated by mixing high-purity sponge Ti (99%) and Ta (99%) by melting and homogenized in a vacuum. It was Cold-worked by a rotary swaging process and further cut to working size. Ti-30Ta alloy substrate surface control was formed by the martensite α” seems to produce a needle-like structure that result in the elastic modulus of 69 GPa and a tensile strength of the 587 MPa with resulting in strength-to-modulus ratio of 8.51 [13]. Recent studies have shown the Ti-30Ta alloy substrates (control) to possess the lamellar and needle-like morphology of the α-phase [21, 22], therefore theses results indicate enables strong mechanical properties and improved corrosion resistance [9-13,18,25] SEM images for Group 1 (Figure 1 (a)) were taken at 25000X. Figure 1(c) represent the SEM images for group 2 were taken at 25000X. The amorphous sodium tantalate layer was observed on Ti-30Ta alloy after alkaline treatment with 1.5 M at 60°C for 24 hrs and heat treatment 300°C for 1 hr. The images show microporous structures with sizes within 50 - 100 nm [47] with cracks uniforms covering all surface [15-17,35-37]. The substrate have different coefficient of expansion that the layer amorphous sodium tantalate formed that results in cracks during the dry in heat treatment [35,48]. Figure 1(e) shows SEM images of Group 3 taken at 25000X with long and confluent particles. The image identified presence of apatite formed on Ti-30Ta alloy substrate after immersion in SBF for 24 hrs covered all substrate. The tantalum metal with alkaline and heat treatment and exposed to SBFx5 shows that the sodium tantalate exchanges its Na’ ion with H2O+ ion in the SBF in order to outcome Ta-OH groups. The nucleation of heterogeneous apatite was formed by the Ta-OH groups. Figure 1(g) shows SEM images for Group 4 taken at 50000X. The results show unorganized nanotubes with 100 nm of the diameter which are not aligned. Figure 1(i) shows SEM images for Group 5 taken at 50000X. The results show formation of nanoscale spheroids with a wavy surface architecture. Further, the results indicate that the ion etching did not significantly alter the surface. It seems that etching at 1200 eV was not enough to significantly change the surface topography [33]. EDS spectra for Group 1 (Figure 1(b)), Group 4 (Figure 1(h)) and Group 5 (Figure 1(j)) showed peaks for titanium and tantalum. Group 2 (Figure 1(d)) showed peaks for titanium, tantalum and oxygen. Group 3 (Figure 1(f)) showed peaks of titanium, tantalum, calcium and sodium. Due to gold coating layer on the surface the phosphorous was not detected by EDS since the phosphorous and gold peaks overlap. Figure 2 shows contact angle measurements on different groups. The results indicate following order of surface hydrophilicity (Figure 2):

Group 4 > Group 2 > Group 3 > Group 1 > Group 5

This behavior is extremely important since cell and bacterial adhesion, protein adsorption, platelet adhesion and activation and blood coagulation may be affected. The materials for biomedical application need to be more hydrophilic since they have higher surface energy which is desirable for biological interaction [43,45,49].

4. Conclusion

In this study, the Ti-30Ta alloy substrates were modified chemically as well as topographically. The surfaces were modified by treating the substrates with alkaline and heat treatment, alkaline and heat-treatments with SBF-coatings, anodization process and ion beam etching. SEM results show different structure on the surface for each group. EDS spectra identified similarity on Groups 1, 4 and 5. The results presented here indicate for Group 2 an amorphous sodium tantalate hydrogel layer on the substrate surface and for Group 3 the apatite nucleation on substrate surface. The Group 4 shows unorganized and vertically nanotubes and Group 5 shows a little alteration in the topography on the substrate surfaces. Overall the contact angle shows Group 5 the most hydrophobic and Group 4 the most hydrophilic. In conclusion, the study indicates Groups 3 and 4 are the more indicated for biomedical application. The current research is now focused.
Figure 1. SEM images (left) and EDS scans (right) for Ti-30Ta alloy Group 1: Ti-30Ta alloy substrates control ((a), (b)), Group 2: Ti-30Ta alloy substrates with alkaline and heat-treatments ((c), (d)), Group 3: Ti-30Ta alloy substrates with alkaline and heat-treatments with SBF-coatings ((e), (f)) with 25000X magnification. Group 4: Ti-30Ta alloy substrates anodized to form nanotubes ((g), (h)), and Group 5: Ti-30Ta alloy substrates after etch ((i), (j)) taking in 5000× magnification.

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